

Using microtomography in geosciences

Ex: Studying fluid and mass transfers in porous geomaterials
with application to CO₂ underground storage, ...

Philippe Gouze

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Philippe Guze (Transport in Porous Media research unit)

- **CNRS** (Governmental Research Institute): \cong 20000 people; all disciplines
 - **Géosciences Montpellier :**
Research covers a large range of topics in geology, geophysics and hydrogeology
 \cong 150 people (1/3 academics, 1/3 CNRS), publications \cong 130/yr, PhD defense \cong 15/yr.
- Based at the **University of Sciences - Montpellier**
 \cong 4500 people (3/5 academics), \cong 16000 students, 7 departments (\cong 90 research units)

To make a long story short:

CO₂ storage (i.e. massive injection of CO₂ in underground reservoirs or aquifers) produces highly aggressive fluids and strong mechanical stresses.

The standard *reservoir* models developed, for instance, for the oil industry cannot handle these conditions.

Many coupled mechanisms taking place in these conditions are still poorly known.

The acceptability (both in term of industrial feasibility and risk assessment) of this technology requires sound predictive models.

The main issues are:

- Identifying all the mass transfer processes
(ex: fluid-rock reactions)
- Measuring parameters for feeding the numerical models
(ex: permeability, dispersivity)
- Determining the functional relationships required for coupling flow, transport and reaction in the numerical models
(ex: porosity-permeability laws)

In this context X-ray microtomography is used to study

Media composition (rock-forming minerals and void distribution)

Media structure *versus* flow properties

Media structure *versus* solute transport & reactions

Structure of fluid-fluid interfaces (multiphase flow)

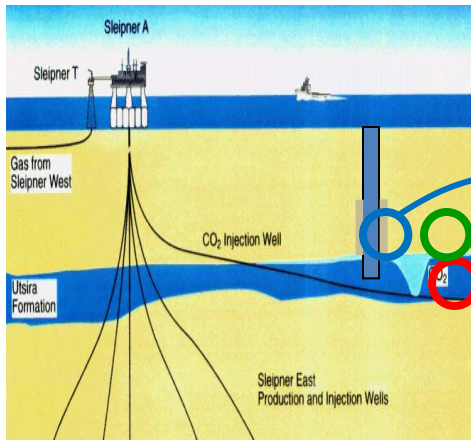
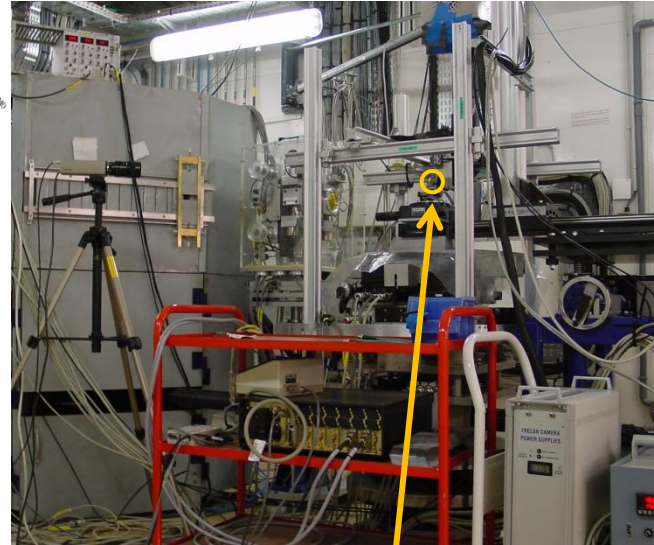
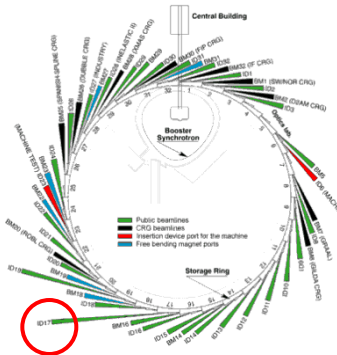
Media structure *versus* mechanical properties

...

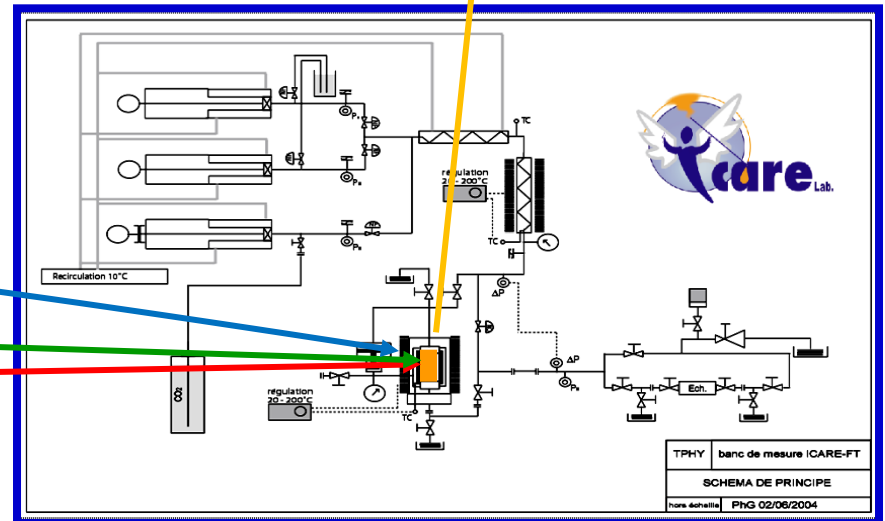
media = porous media or fractured media

= reservoir rocks, caprocks, well cement, ...

Typical study protocol



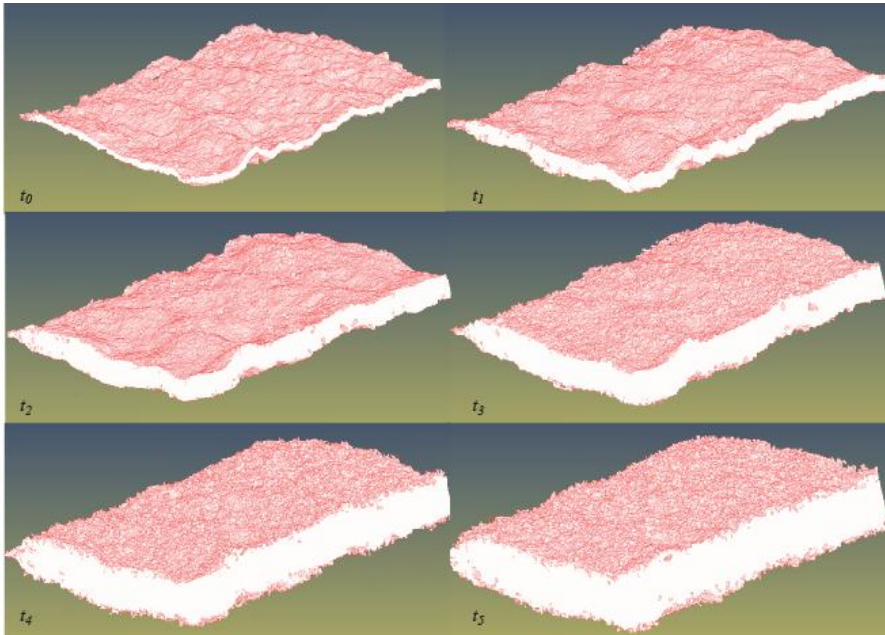
Cement samples
Caprock samples
Reservoir rock samples



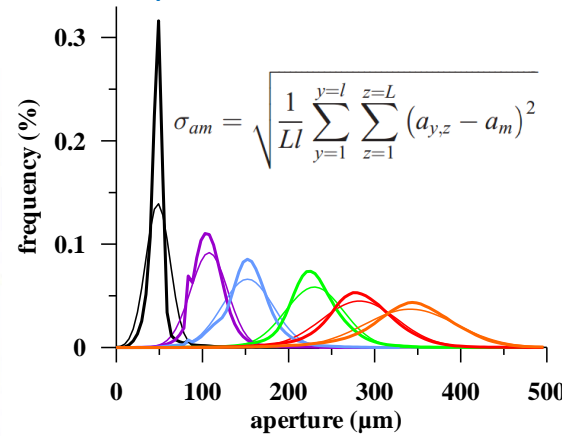
Reproduce *in situ* conditions

fractures

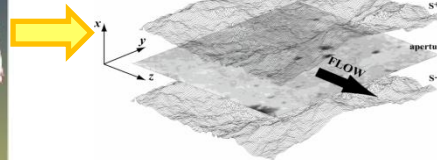
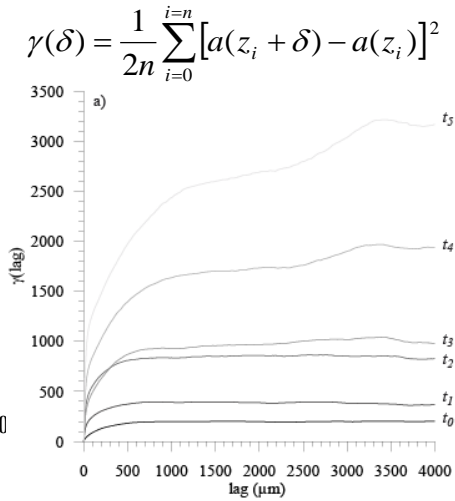
Rough fracture dissolution : geometrical properties *versus* aperture increase



Aperture distribution



Aperture spatial correlation

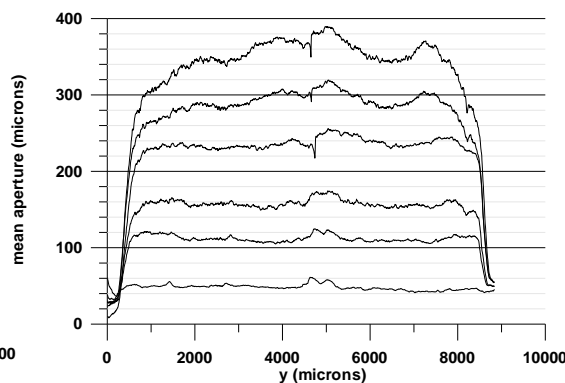
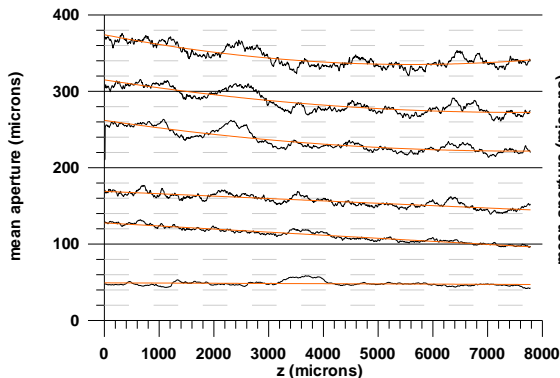


Many other parameters ...

$$\tau = \frac{1}{l} \sum_{y=1}^{y=l} \frac{L'}{L} = \frac{1}{l} \sum_{y=1}^{y=l} \frac{\sum_{z=1}^{z=L} \sqrt{(h_{z+1} - h_z)^2 + \Delta z^2}}{L}$$

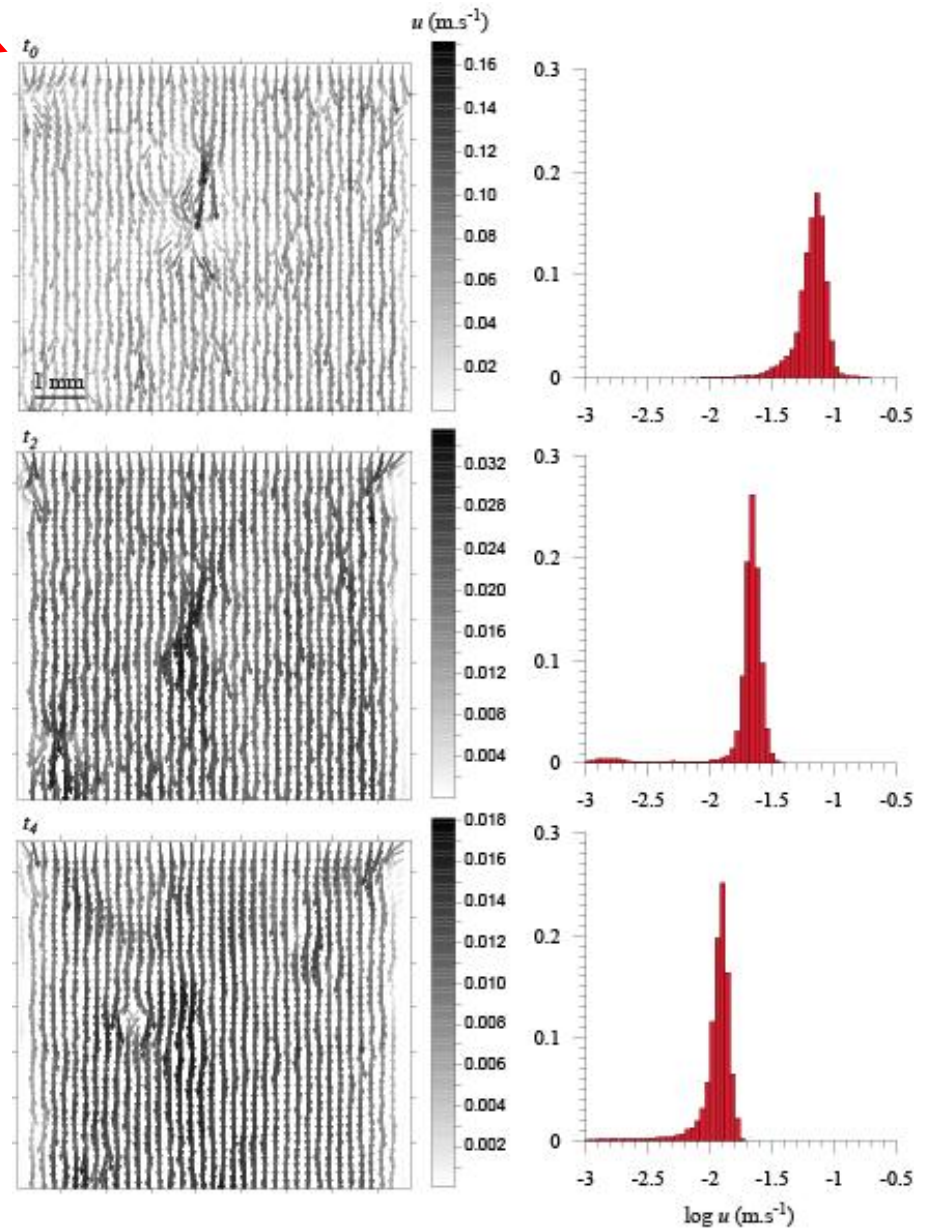
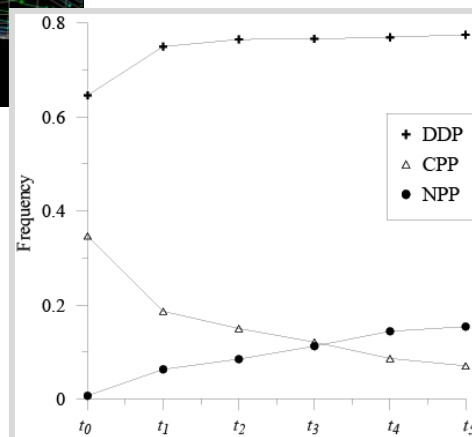
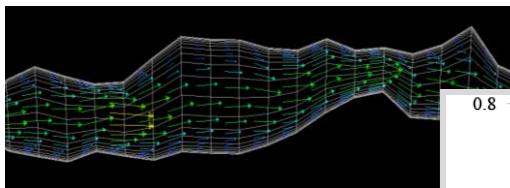
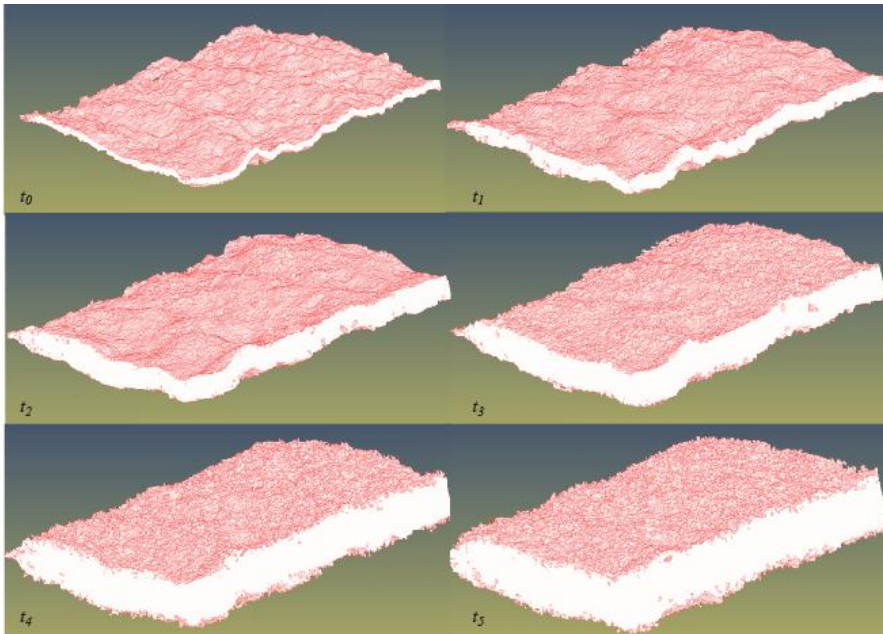
$$\mathfrak{R} = \sqrt{\frac{1}{Ll} \sum_{y=1}^{y=l} \sum_{z=1}^{z=L} (h_{z+1} - h_z)^2}$$

Average aperture // and ⊥ to flow



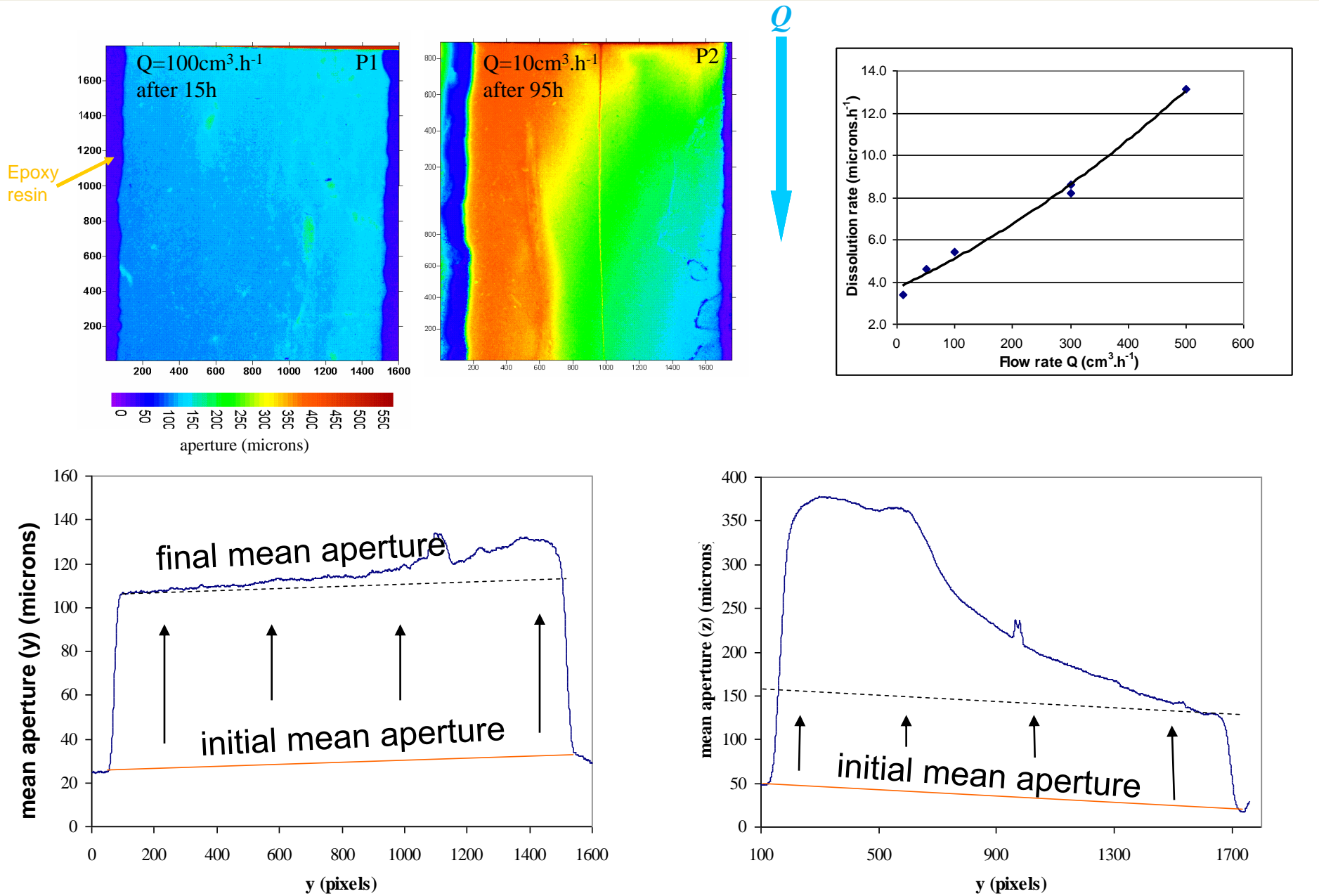
Rough fracture dissolution : flow field properties *versus* aperture increase

Solve Stokes equation

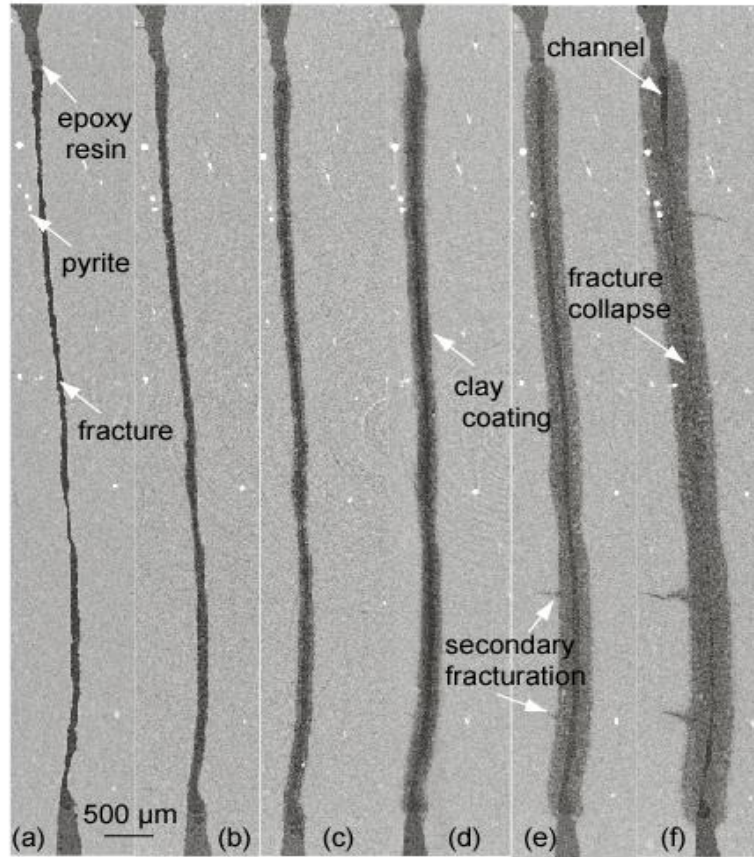


(Noiriel and Gouze, JH, in review)

Planar fracture (sawn) : localization effects and dissolution rate *versus* Q

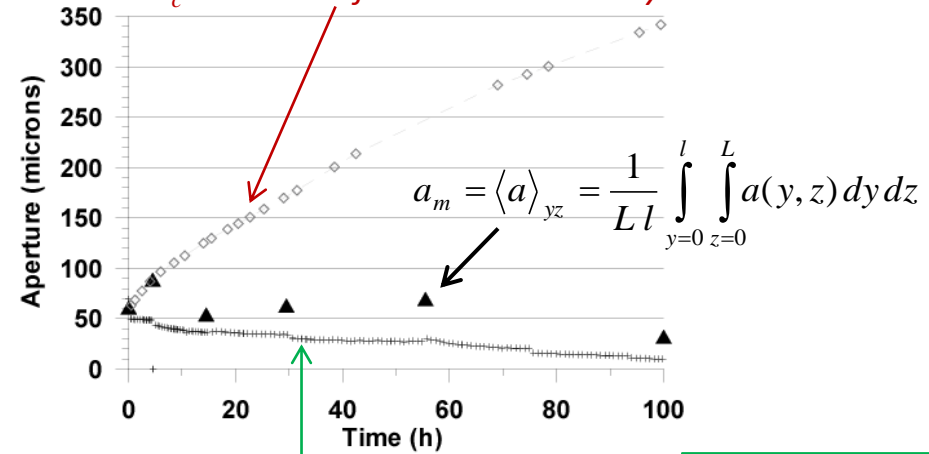


Rough fracture dissolution with coating growth (marl)



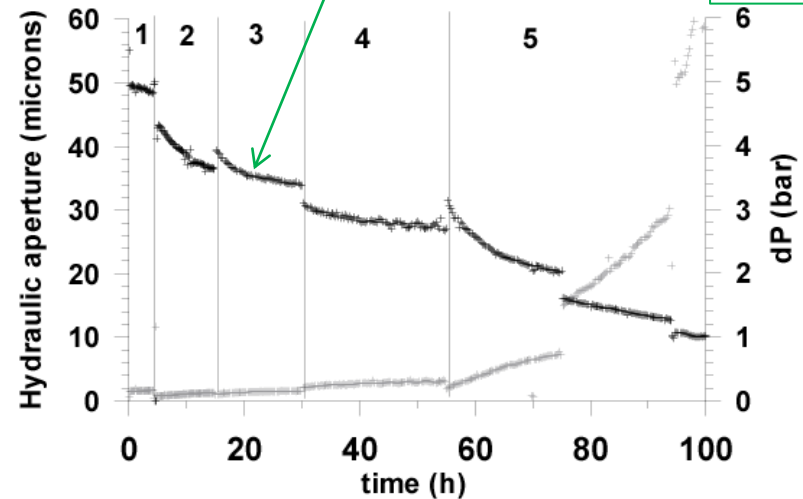
$$a_c = a_0 + \frac{Q}{A_s} \times \left(v_{dol} \int_{t_0}^{t_i} \Delta C_{Mg} dt + v_{cal} \int_{t_0}^{t_i} (\Delta C_{Ca} - \Delta C_{Mg}) dt \right)$$

a_c calculated from chemical analysis

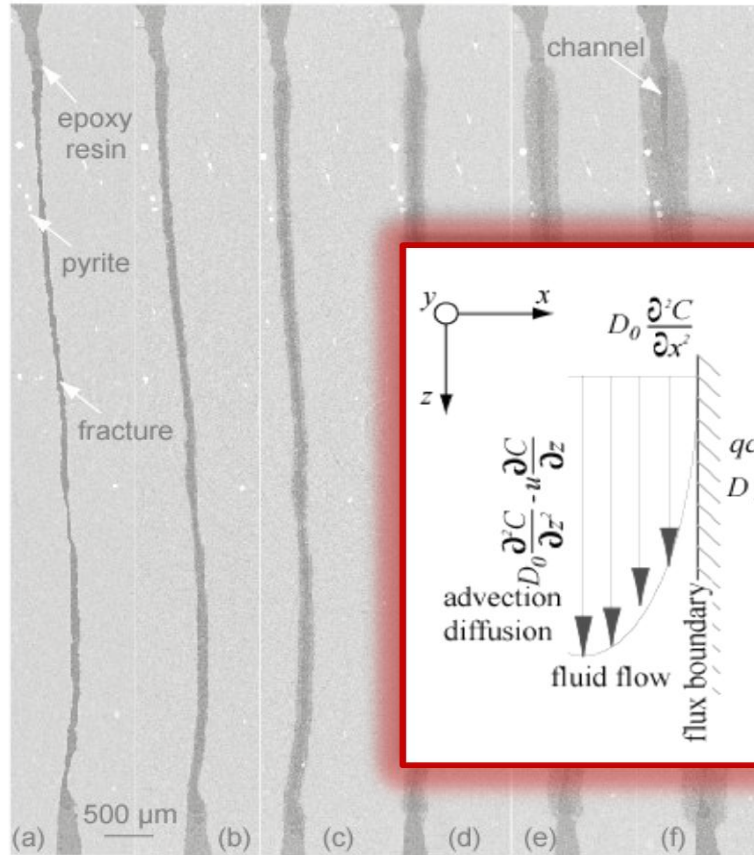


a_h calculated from Darcy's law

$$a_h = \sqrt[3]{\frac{12 \mu L Q}{\Delta P l}}$$

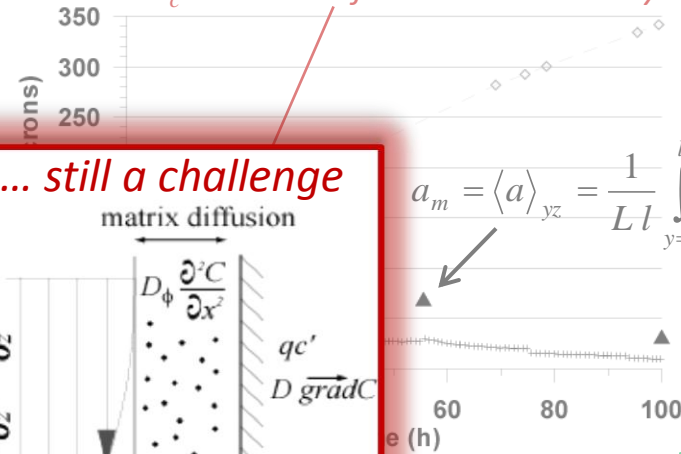


Rough fracture dissolution with coating growth (marl)

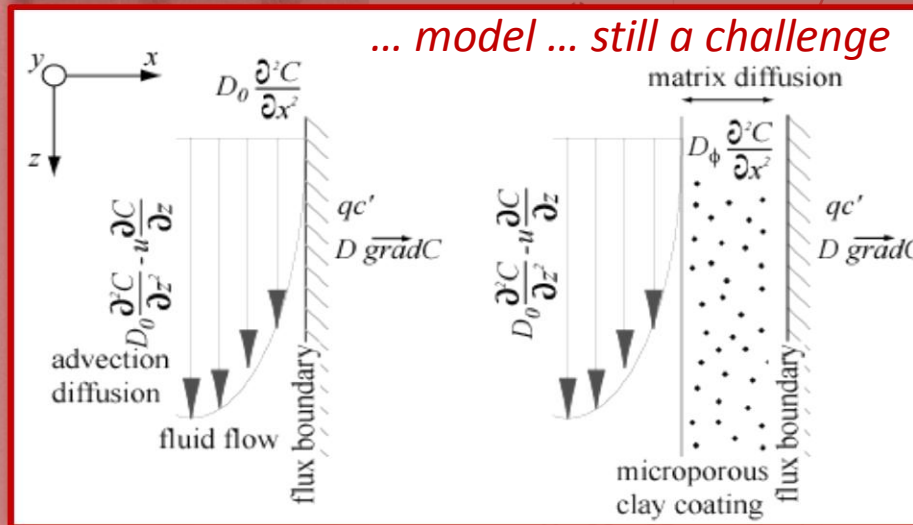


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a_c calculated from chemical analysis



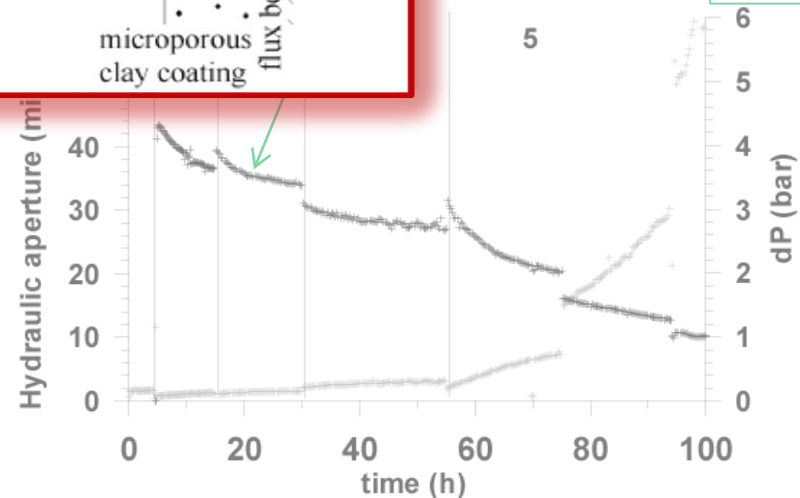
... model ... still a challenge



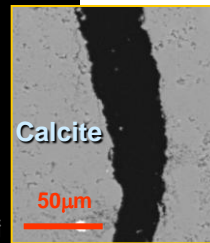
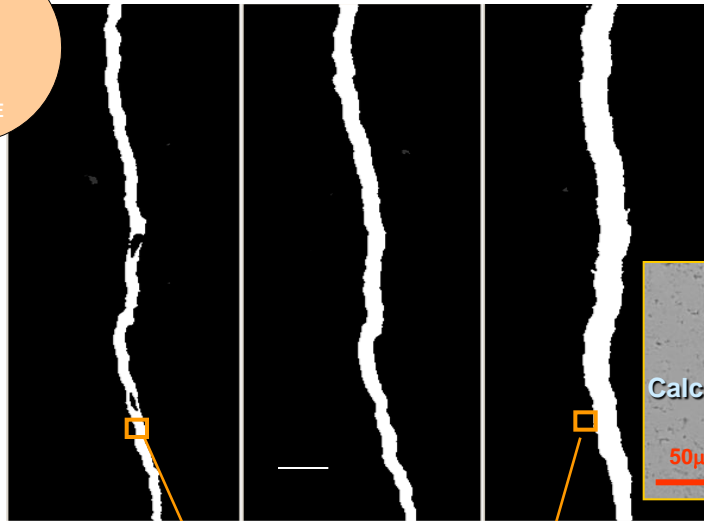
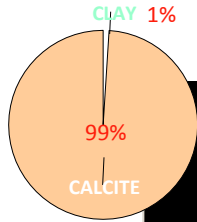
$$a_m = \langle a \rangle_{yz} = \frac{1}{Ll} \int_{y=0}^l \int_{z=0}^L a(y, z) dy dz$$

from Darcy's law

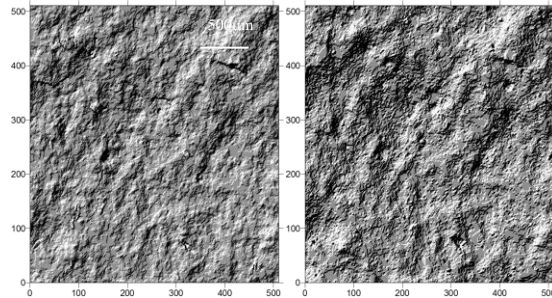
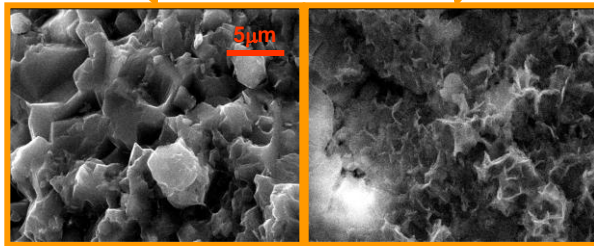
$$a_h = \sqrt[3]{\frac{12 \mu L Q}{\Delta P l}}$$



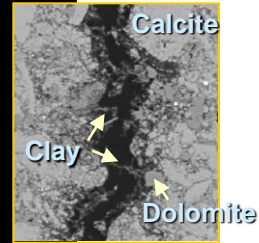
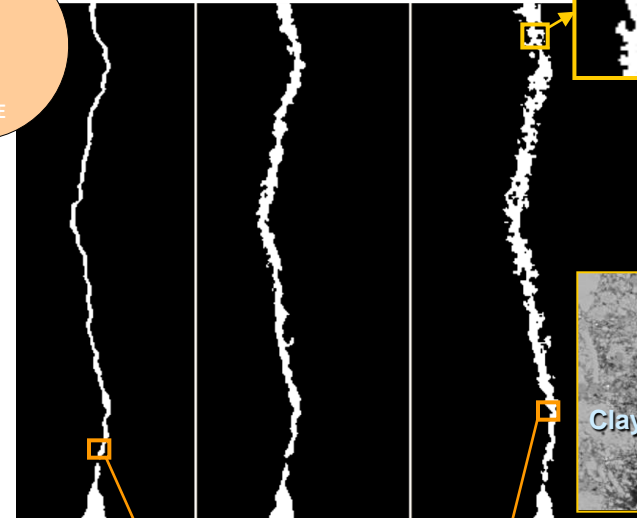
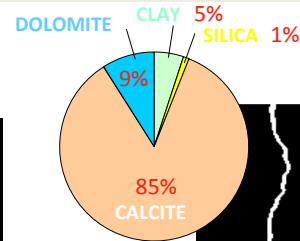
Rough fracture dissolution: effect of differential kinetics



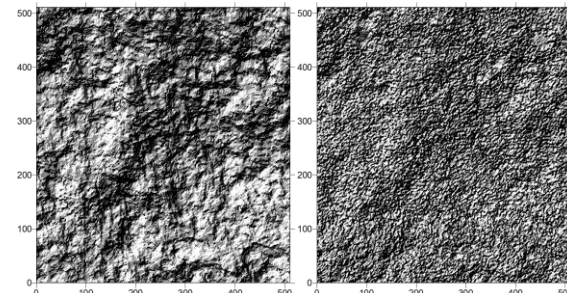
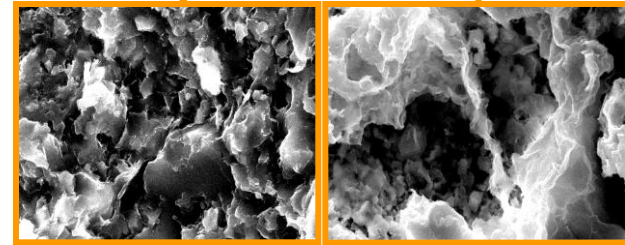
Homogeneous dissolution



Fractal dimension: 2.2 → 2.2

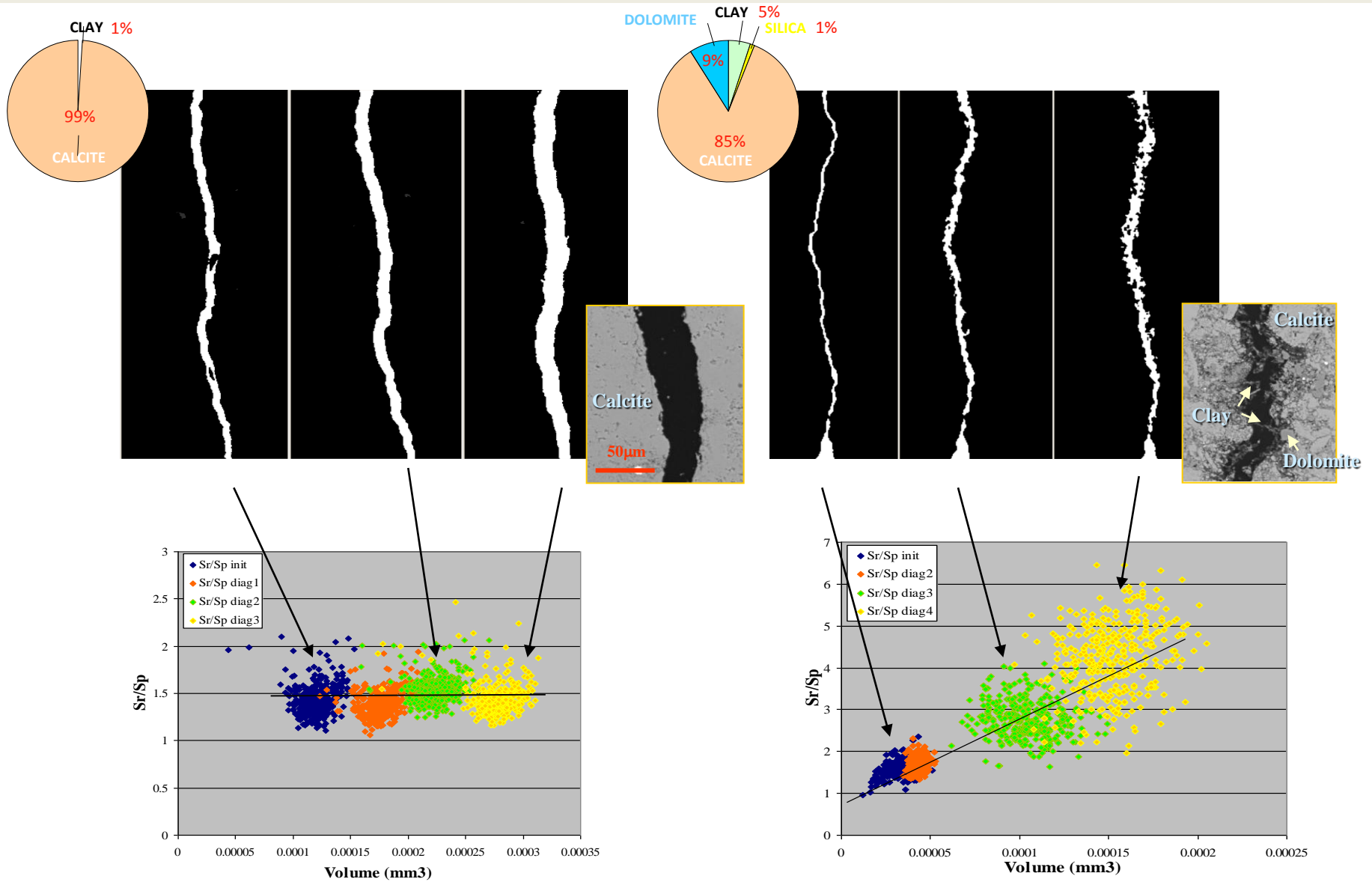


Heterogeneous dissolution

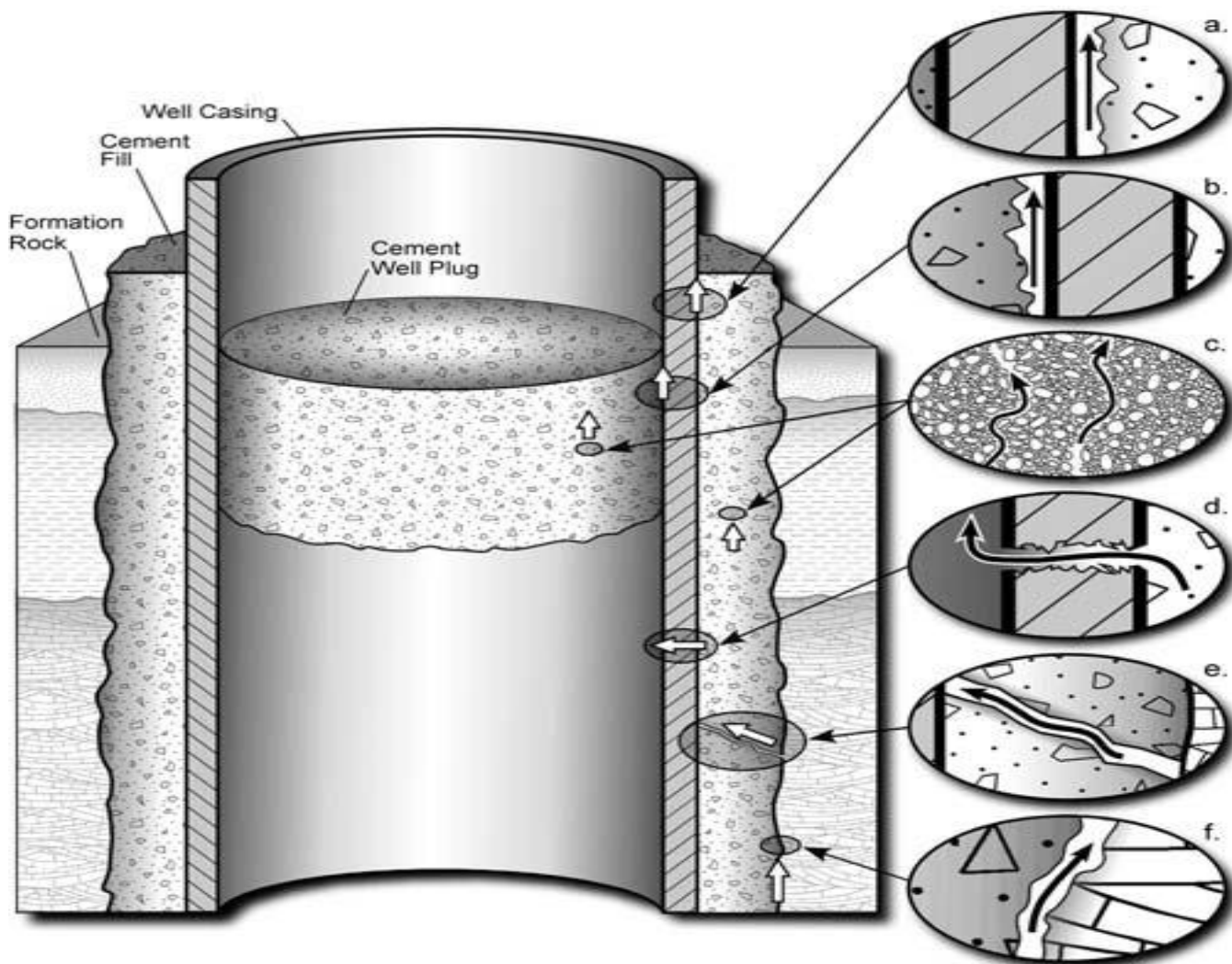


2.2 → 2.4 (Gouze *et al*, GRL, 2003)

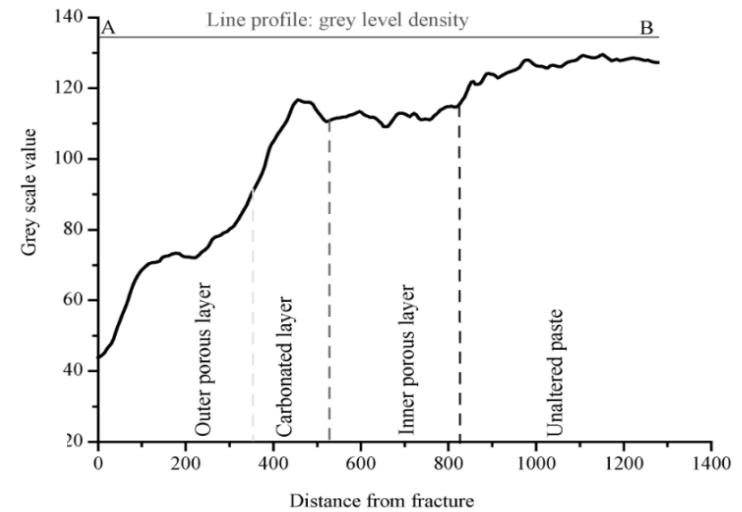
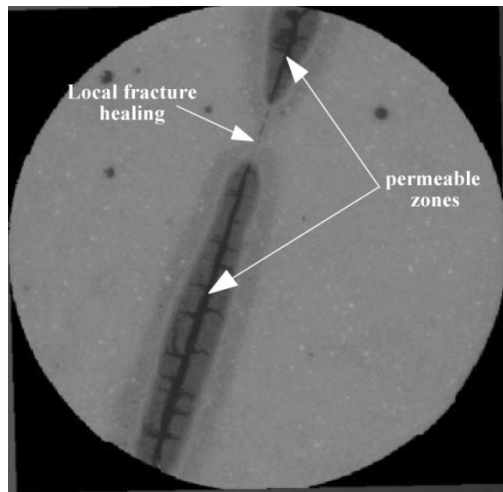
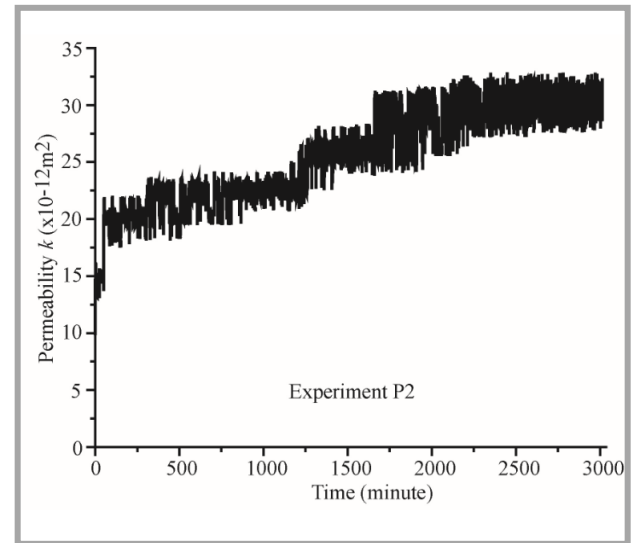
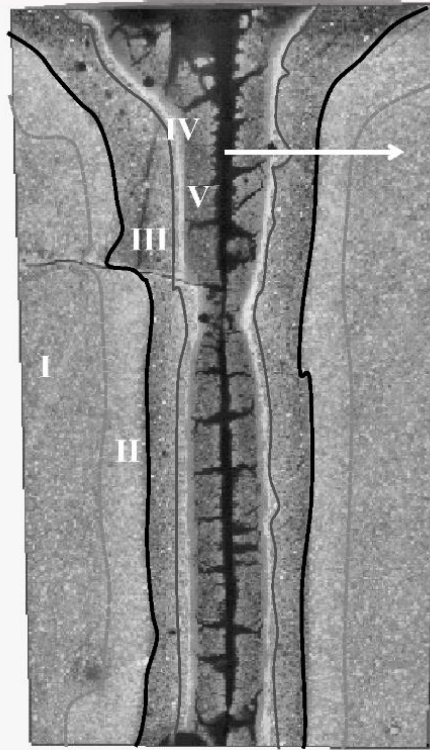
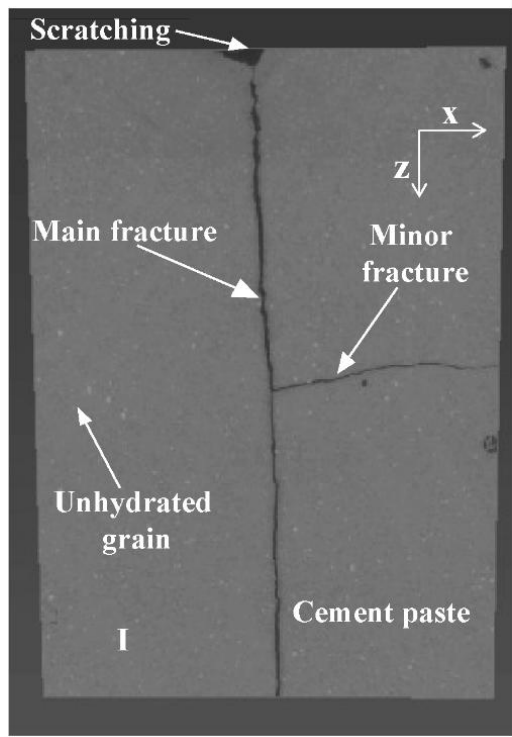
Rough fracture dissolution: effect of differential kinetics



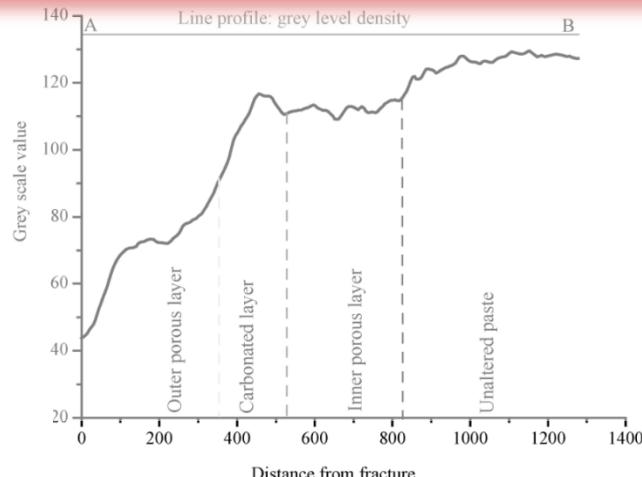
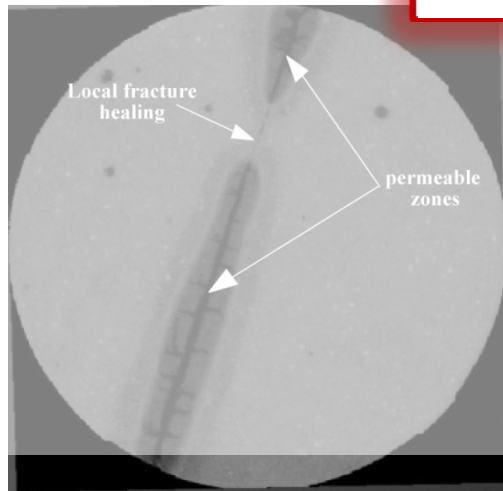
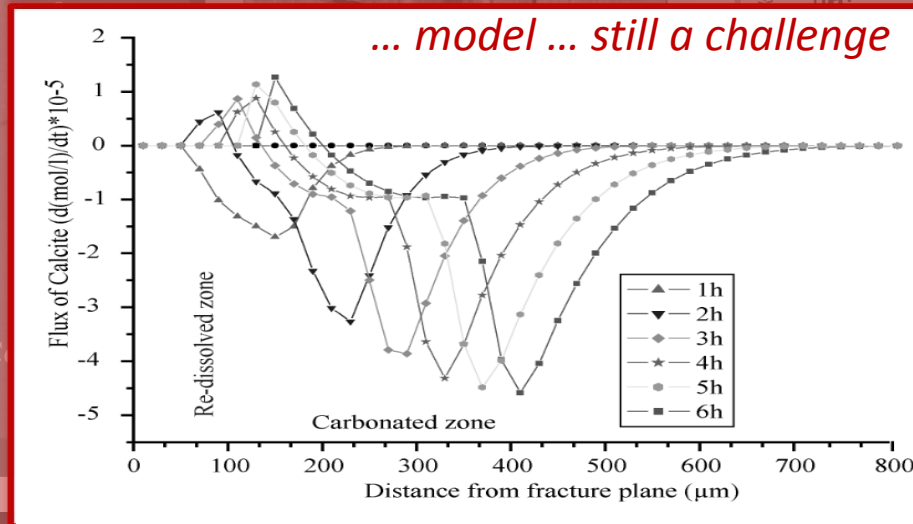
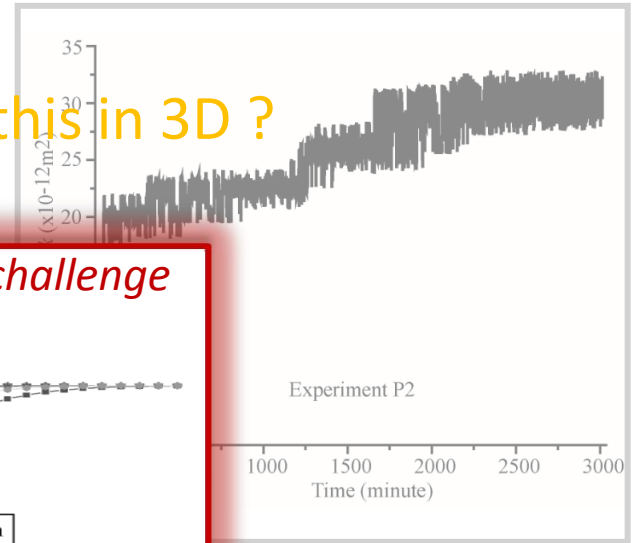
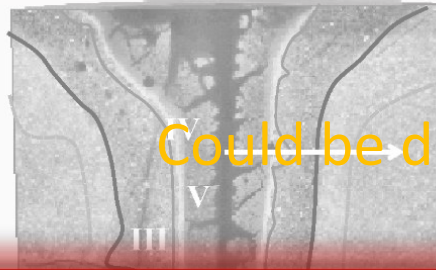
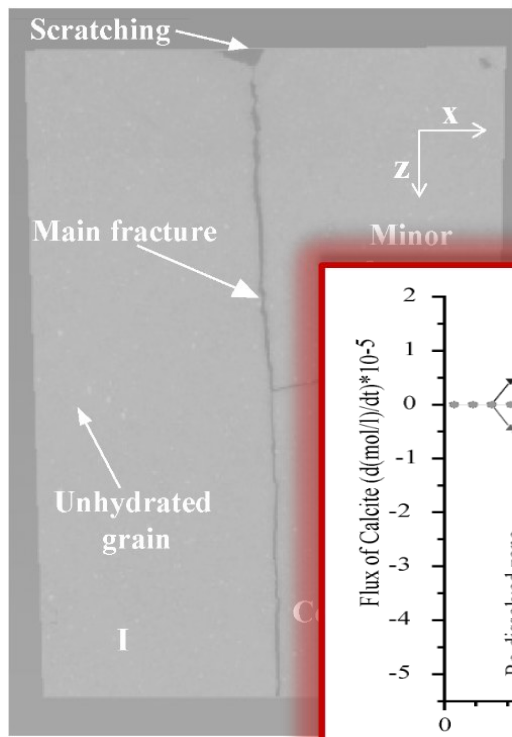
Alteration of fractured class G well cements



Alteration of fractured class G well cements



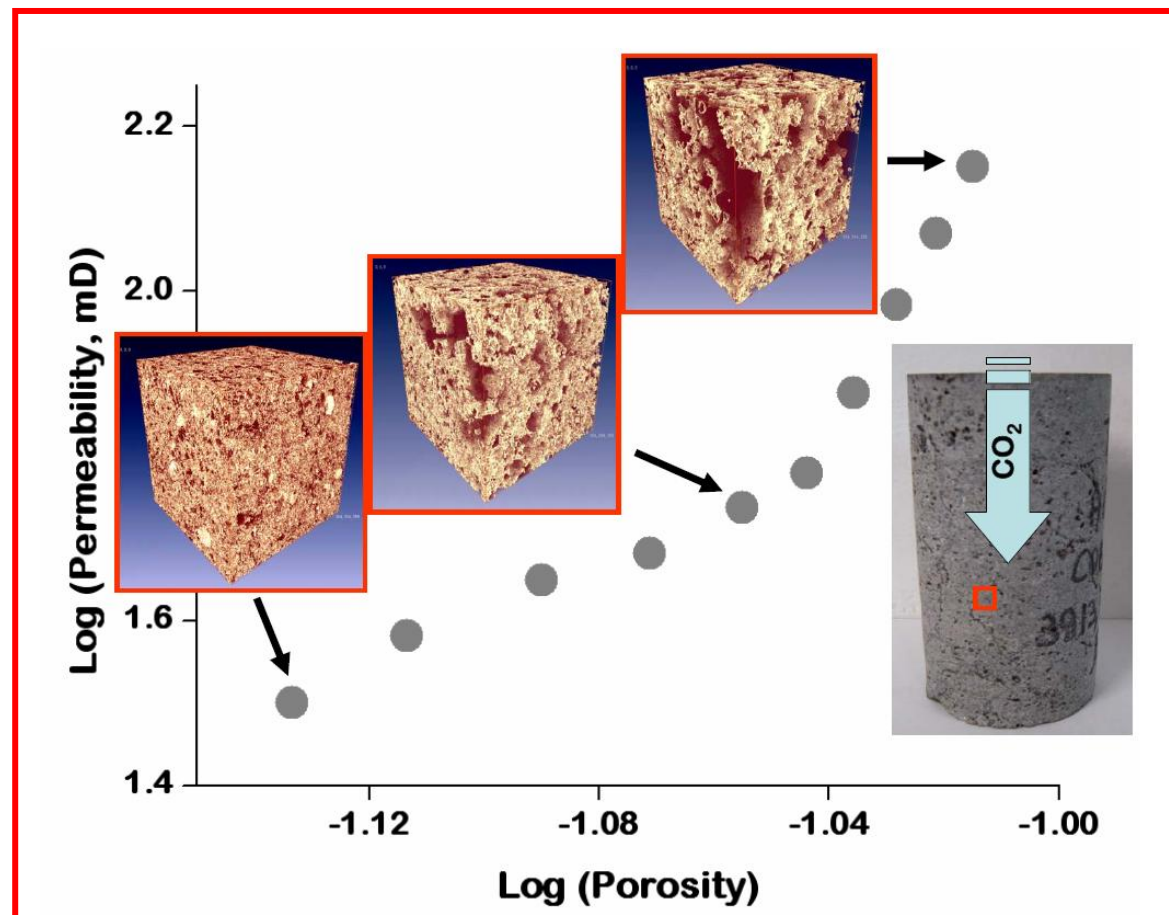
Alteration of fractured class G well cements



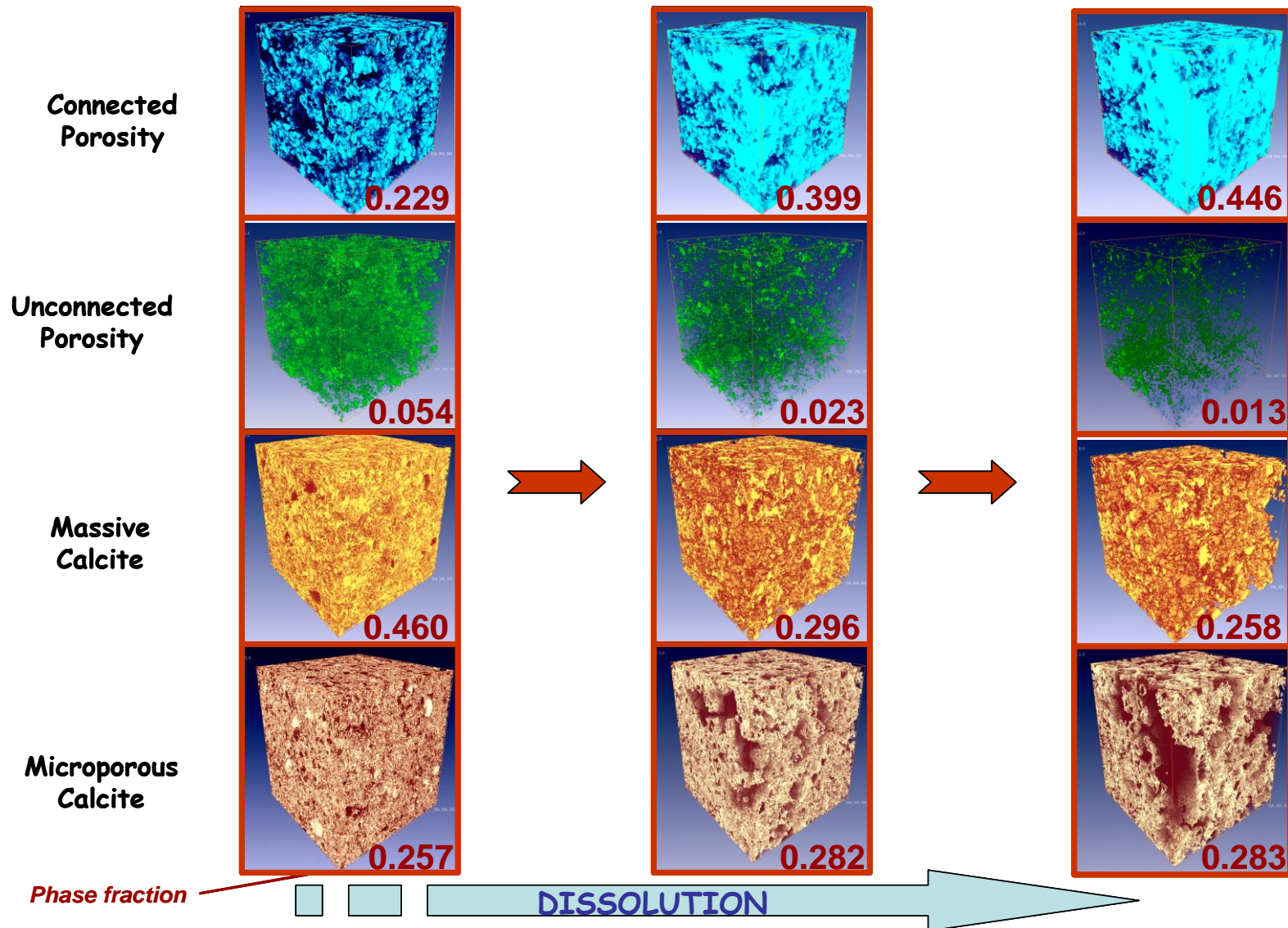
Porous media

Characterization of the dissolution/precipitation processes in the reservoir

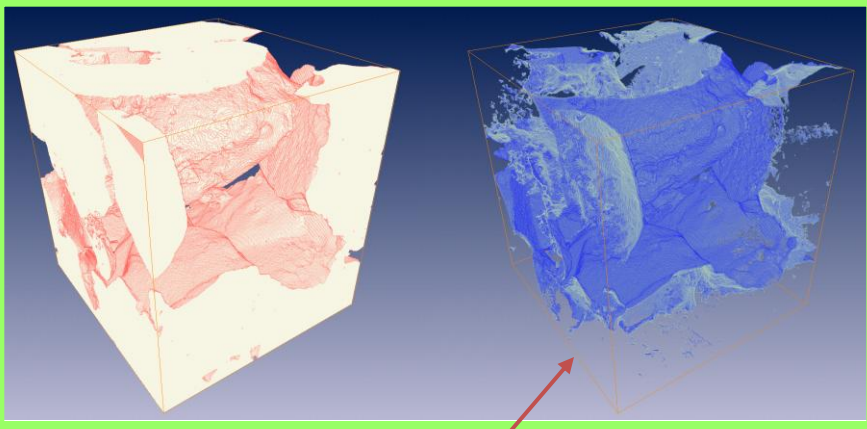
Main issue for model = poro-perm relationship



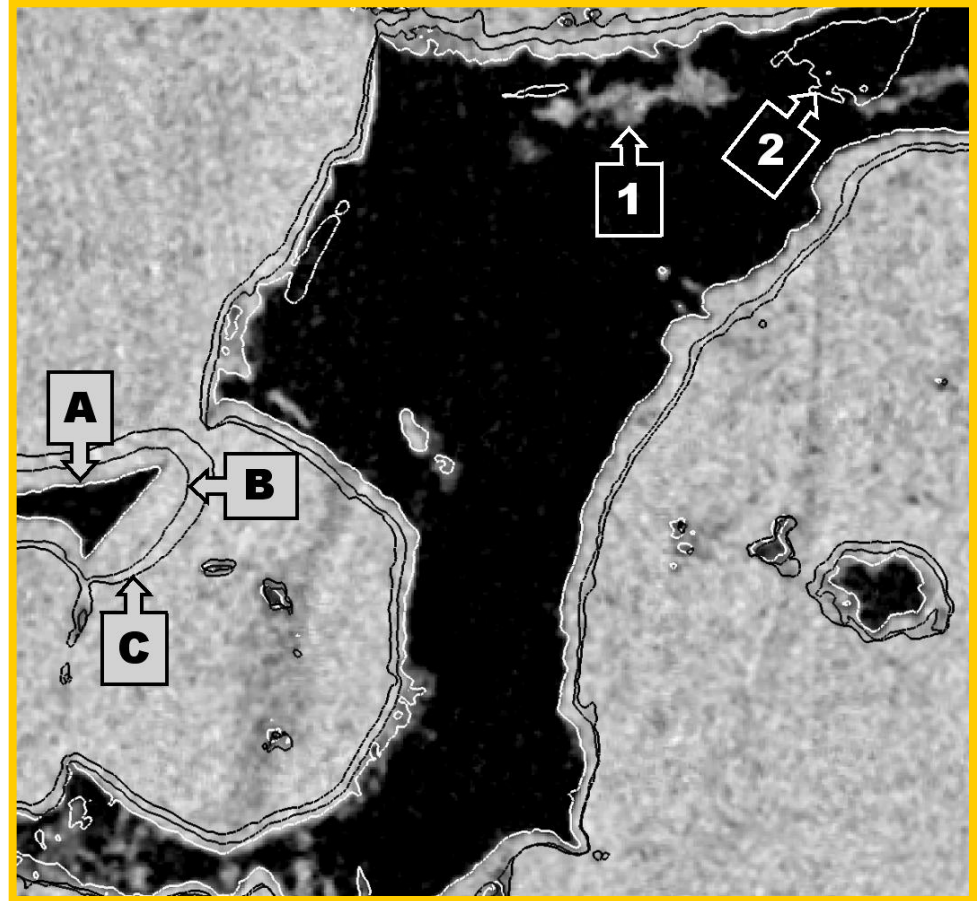
Characterization of the dissolution/precipitation processes in the reservoir



Characterization of the dissolution/precipitation processes in the reservoir



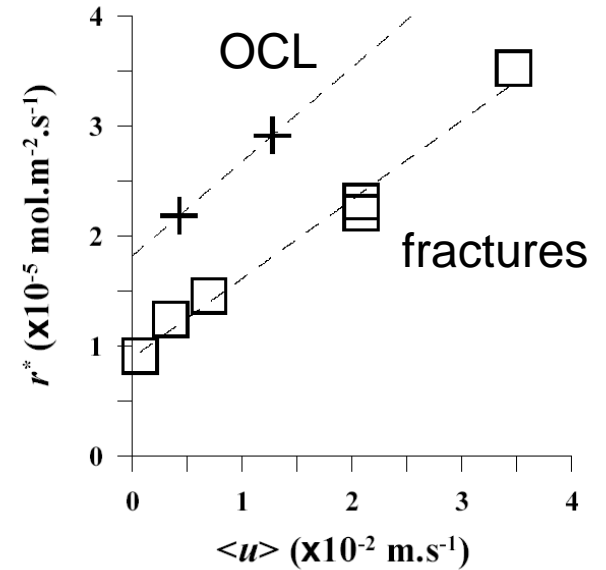
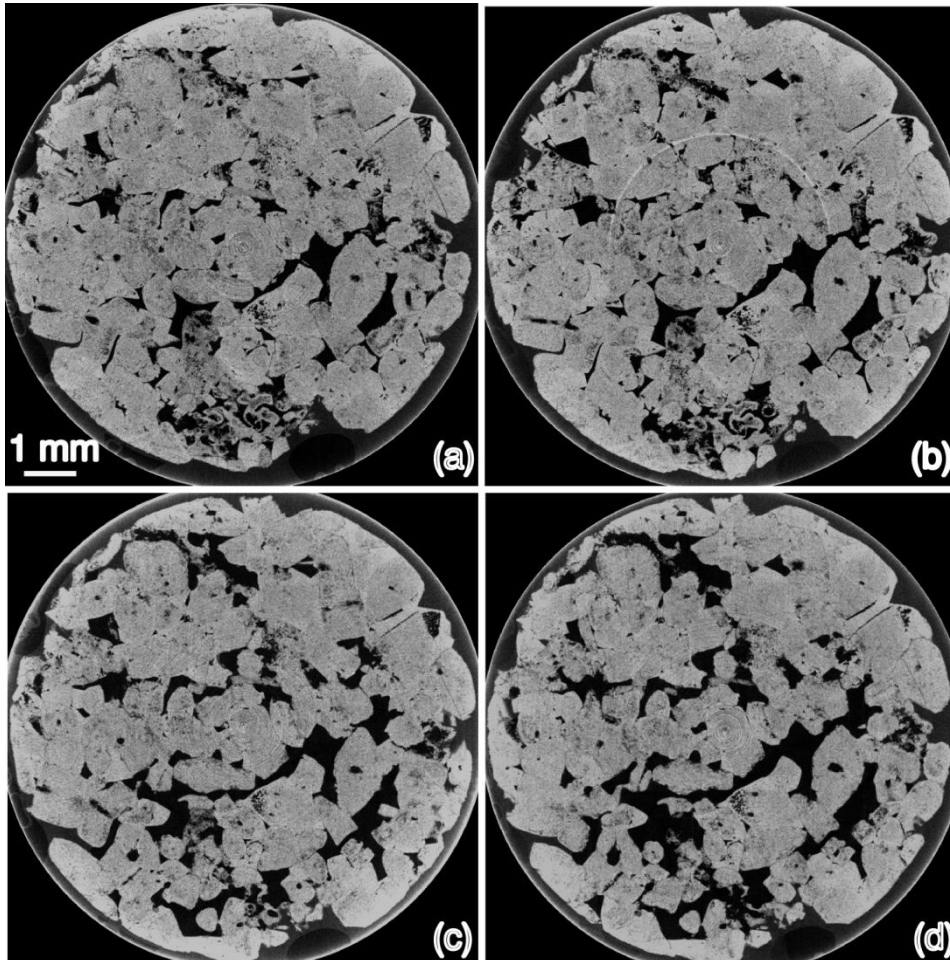
Cumulative volume of rock dissolved at the end of the experiment



Characterization of the dissolution/precipitation processes in the reservoir ...

and measuring effective dissolution rate and kinetics parameters

Oxfordian crinoïdal limestone (OCL)

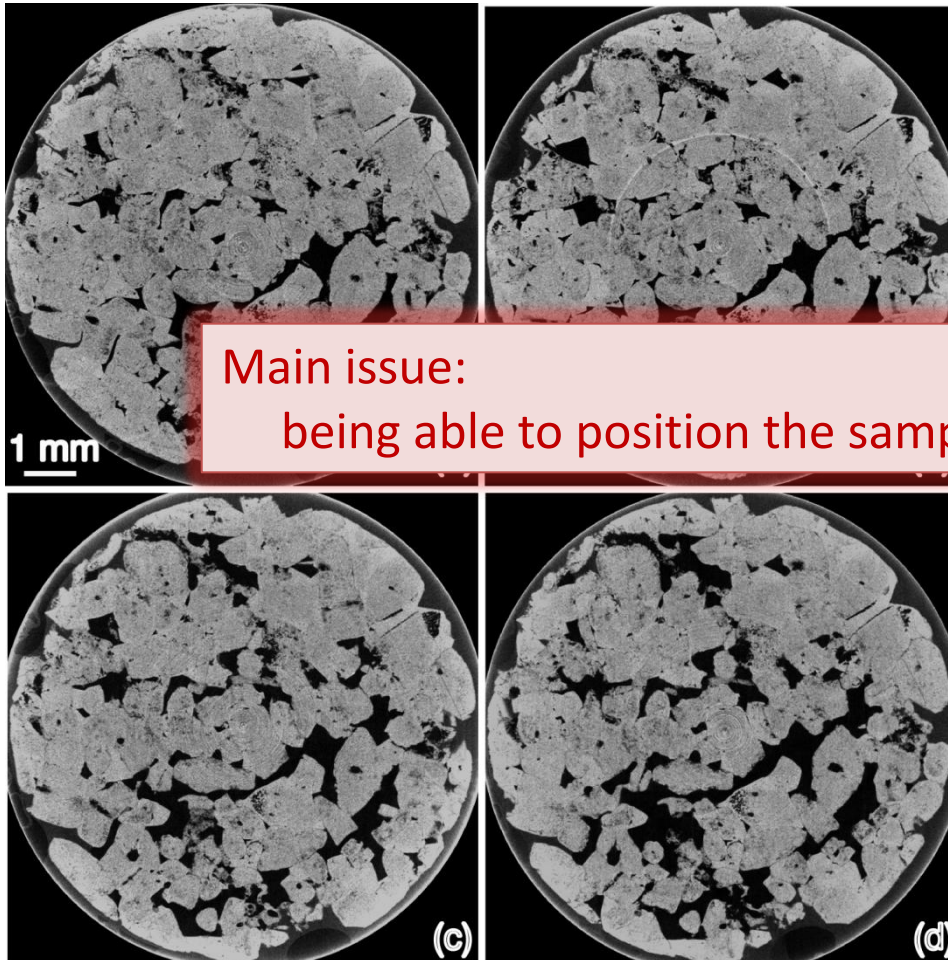


(Noiriel *et al.*, GRL, 2004)

Characterization of the dissolution/precipitation processes in the reservoir ...

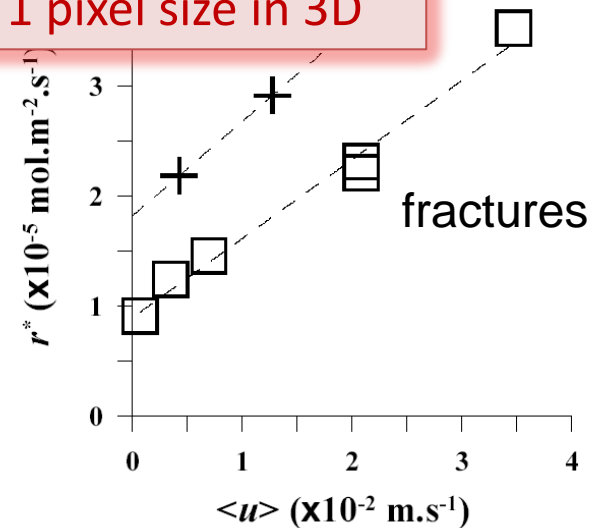
and measuring effective dissolution rate and kinetics parameters

Oxfordian crinoidal limestone (OCL)



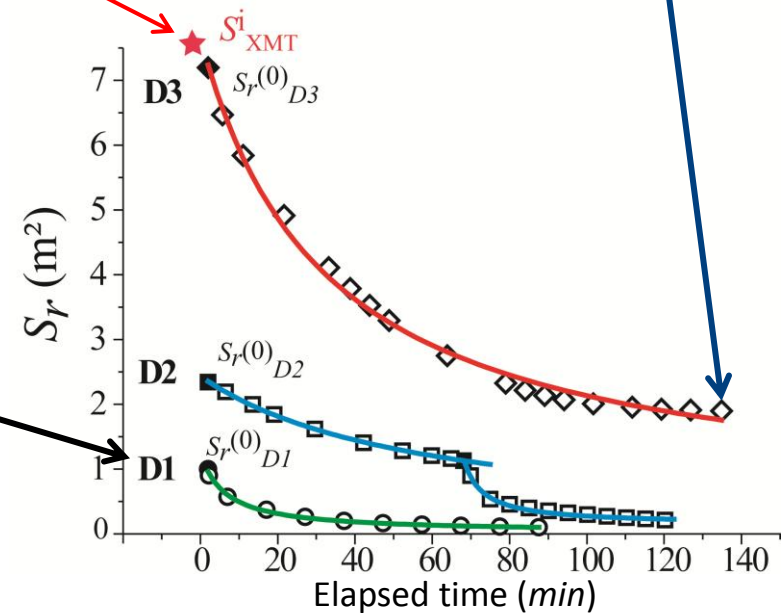
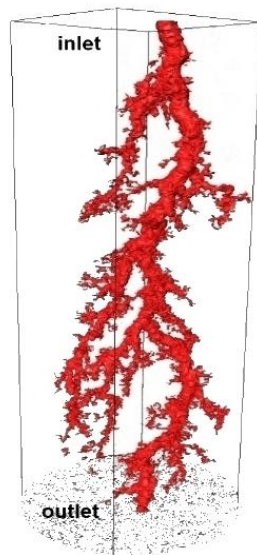
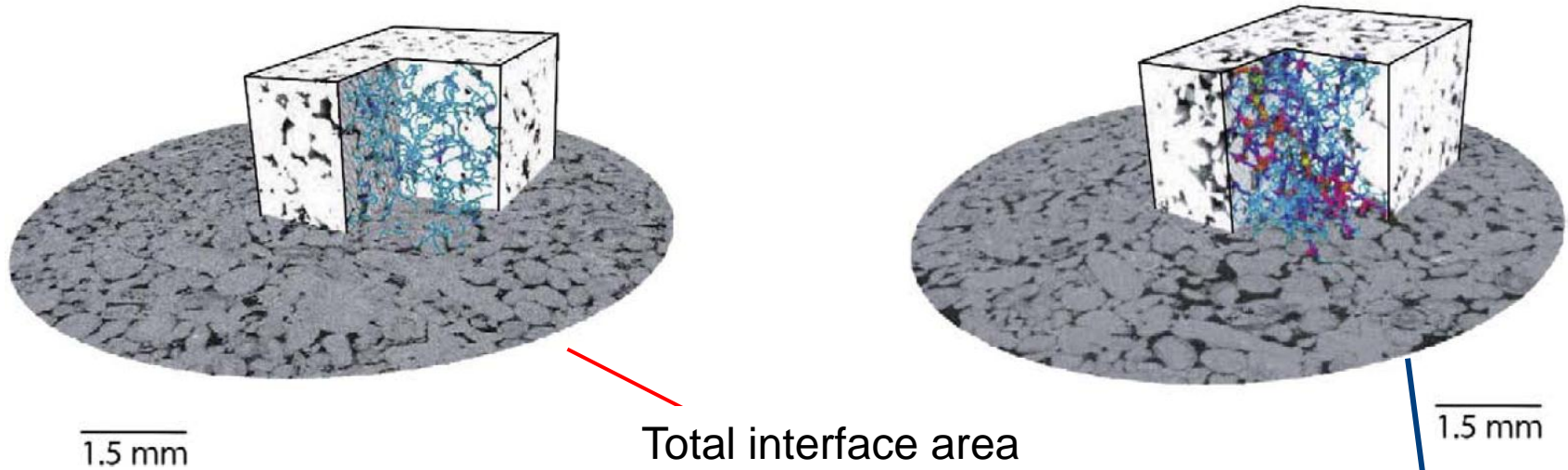
Main issue:

being able to position the sample at ± 1 pixel size in 3D

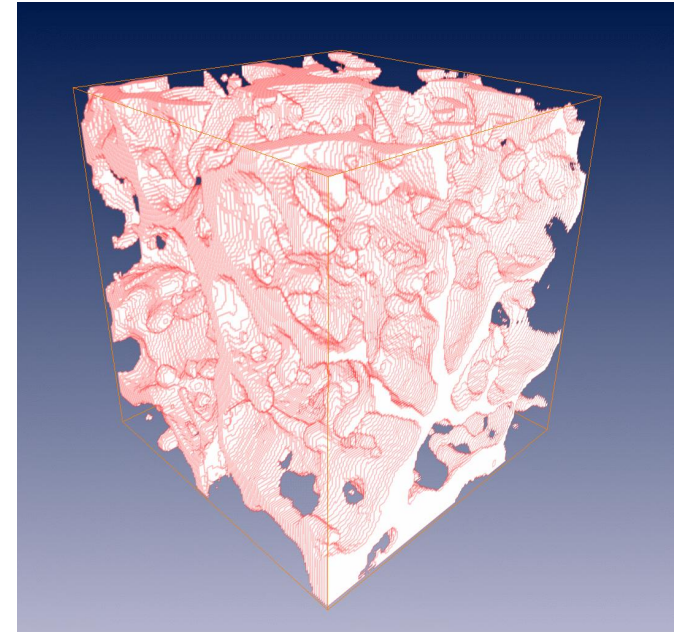
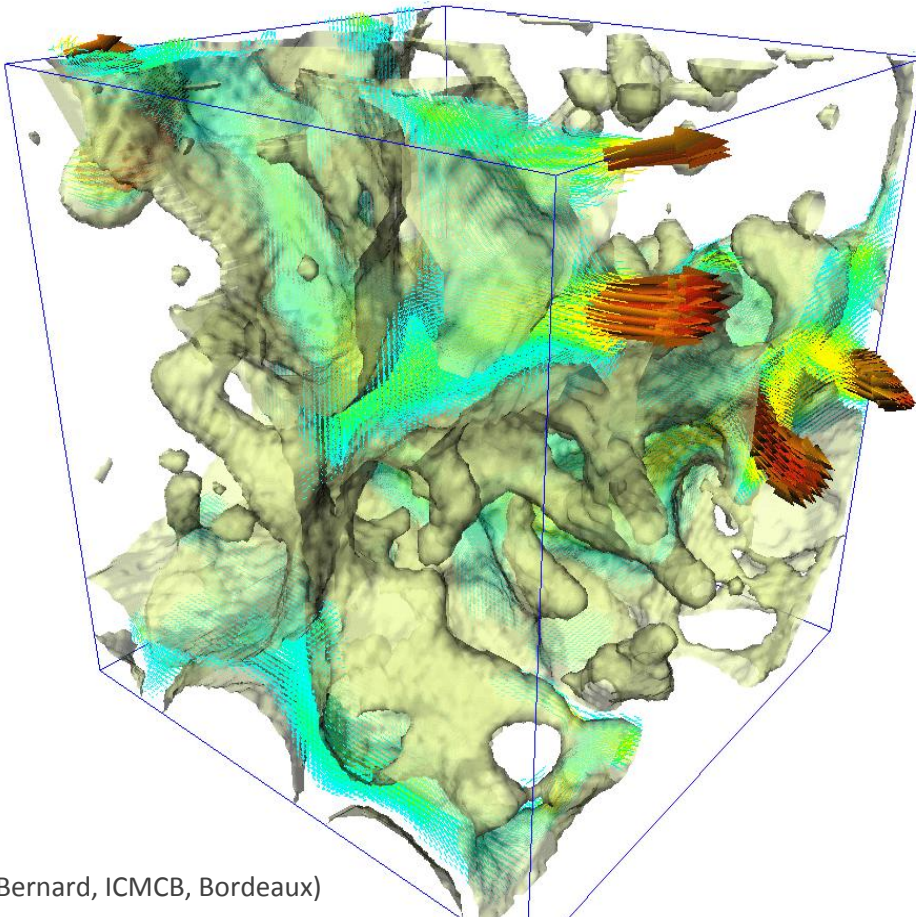


Characterization of the dissolution/precipitation processes in the reservoir ...

and measuring effective reactive surface area



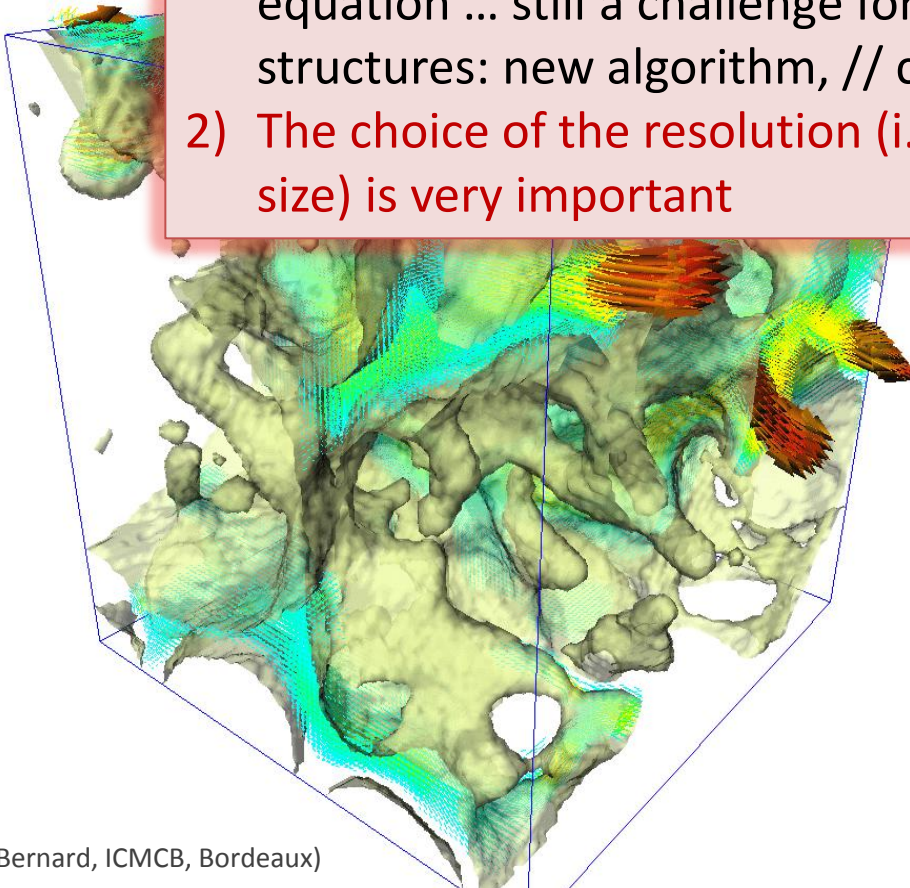
Compute permeability tensor



$$K_{ij} = \begin{vmatrix} 1.216 & 4.77 \cdot 10^{-3} & 1.90 \cdot 10^{-2} \\ 4.78 \cdot 10^{-3} & 1.227 & -0.173 \\ 1.90 \cdot 10^{-2} & -0.173 & 1.046 \end{vmatrix}$$

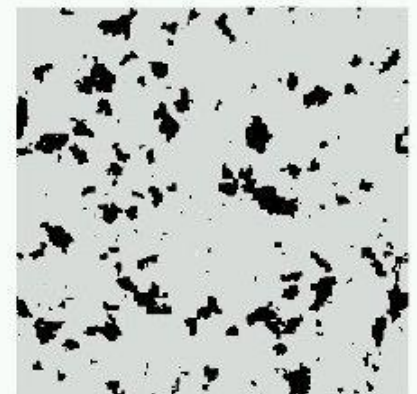
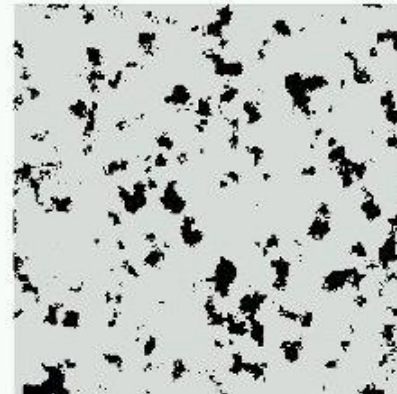
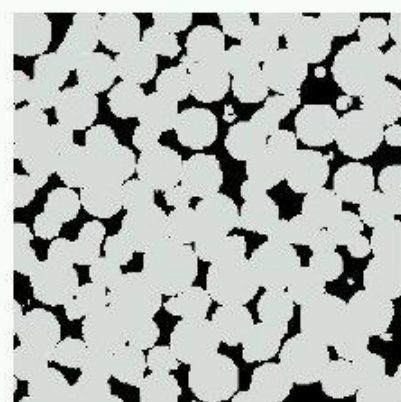
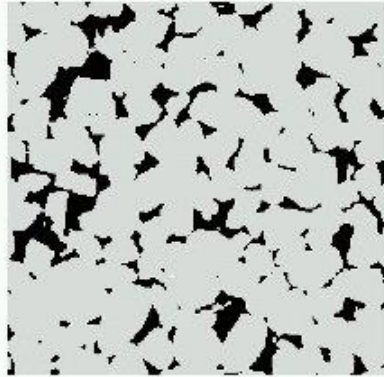
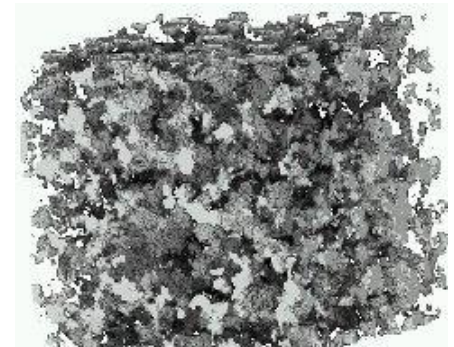
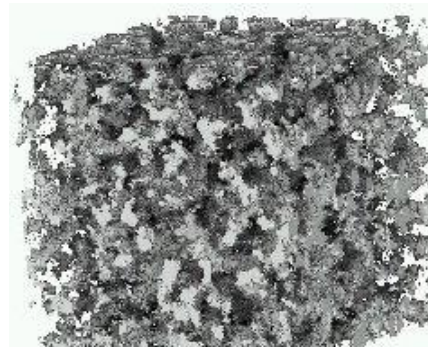
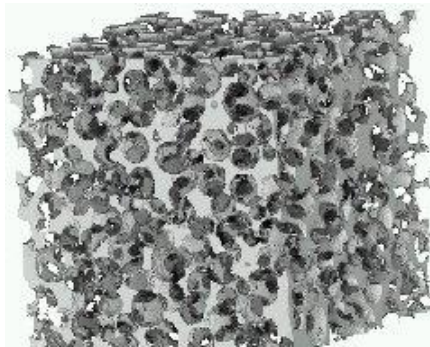
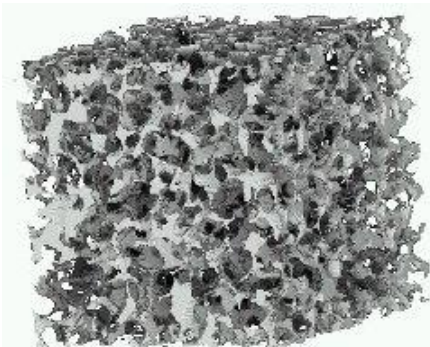
Compute permeability tensor

- 1) Meshing and then compute the Stokes equation ... still a challenge for “large” size structures: new algorithm, // computing, ...
- 2) The choice of the resolution (i.e. the sample size) is very important



$$K_{ij} = \begin{vmatrix} 1.216 & 4.77 \cdot 10^{-3} & 1.90 \cdot 10^{-2} \\ 4.78 \cdot 10^{-3} & 1.227 & -0.173 \\ 1.90 \cdot 10^{-2} & -0.173 & 1.046 \end{vmatrix}$$

Use properties for generating statistically equivalent porous media



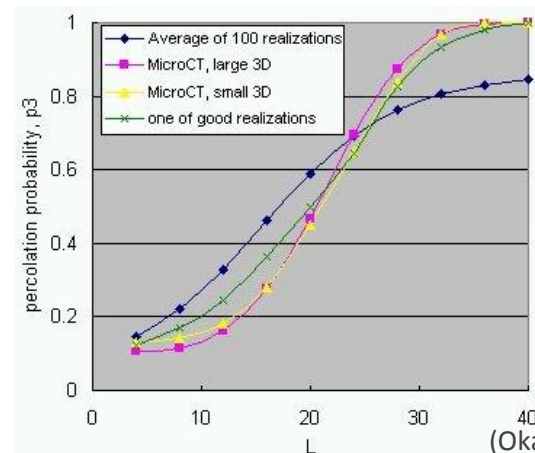
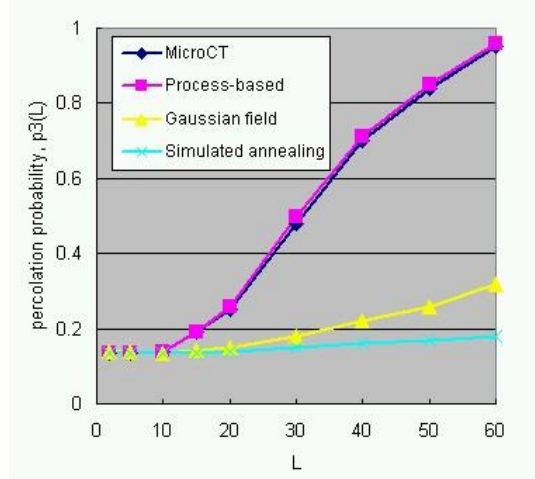
MicroCT

Process-based

Gaussian-field

Simulated Annealing

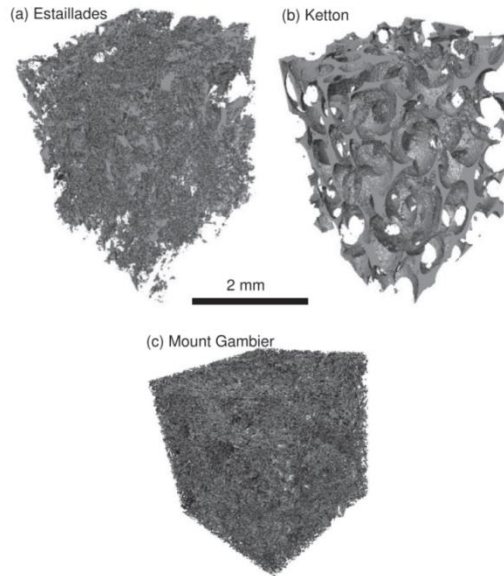
(Biswal et al., Phys. A, 1999)



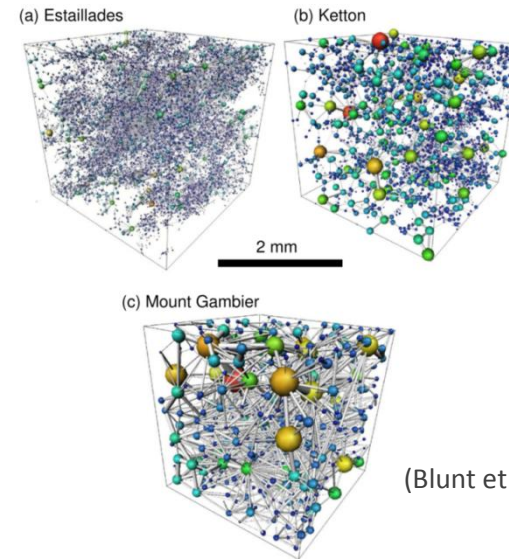
(Okabe and Blunt, PS&E., 2005)

Construction of pore network models

Objective: simplifying computations for flow and transport by using much simple geometries.



Micro CT



(Blunt et al., AWR, 2005)

Pore networks extracted from the images

Construction of pore network models

Objective: simplifying computations for flow and transport by using much simple geometries.



Simplifying computation for flow and transport?

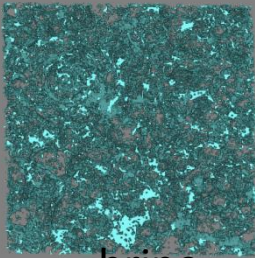
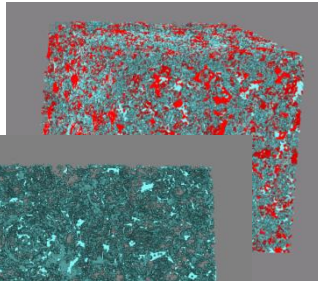
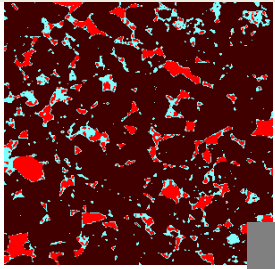
yes if

- 1) fully reversible skeletonization
- 2) structure characterization over
5 to 6 order of magnitude

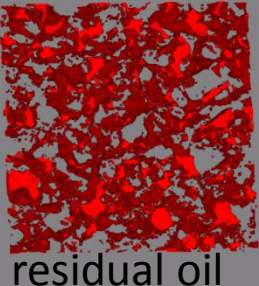
ex: sample 10^{-2} m \Rightarrow tomo @ 5×10^{-6} m ,
then, zoom-in @ 5×10^{-7} m
or, zoom-in @ 5×10^{-8} m !)

Multiphase flow

CT-scan



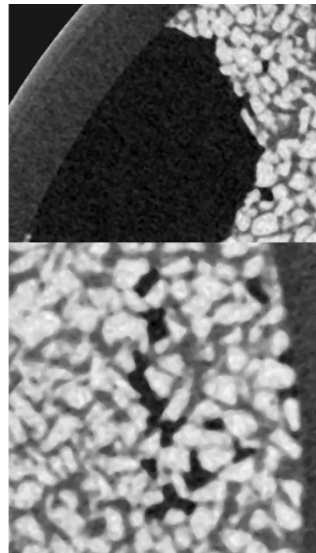
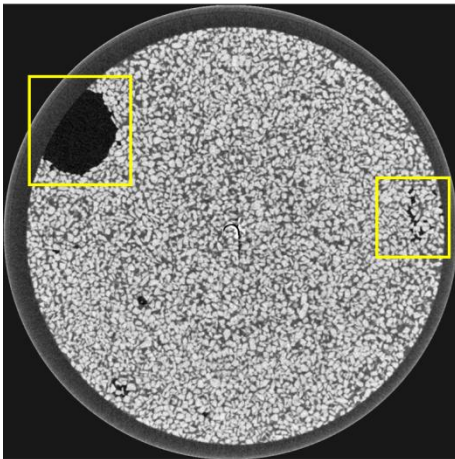
brine



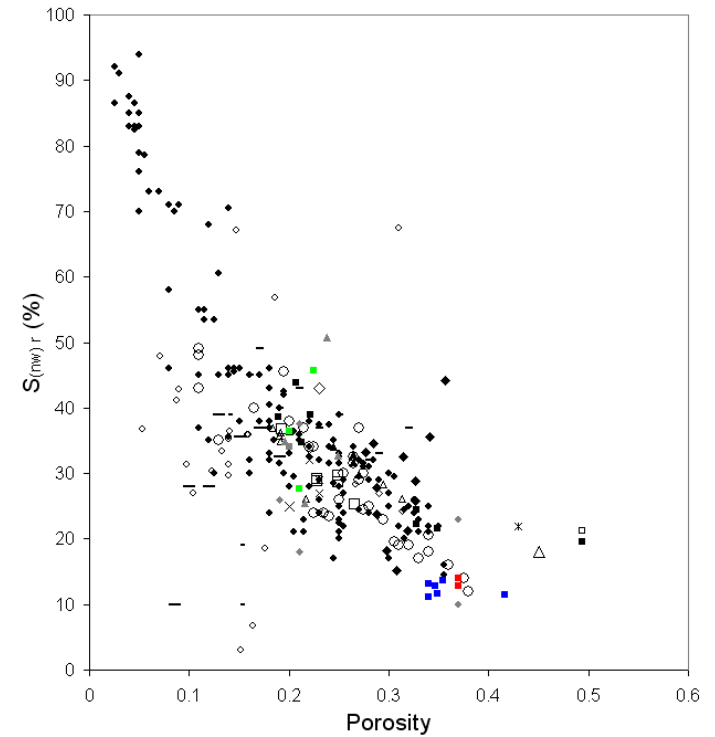
residual oil

(Pentland et al., GRL, 2011)

Synchrotron (ESRF)

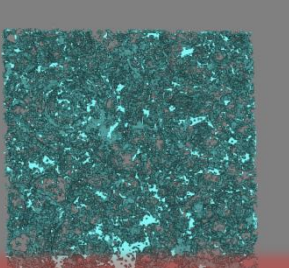
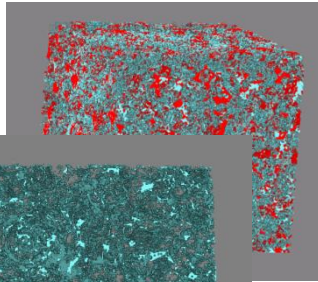
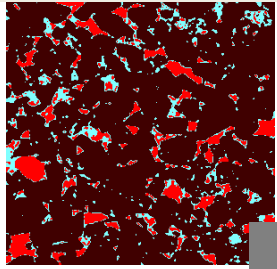


Objective: measure residual saturation
& relative permeability

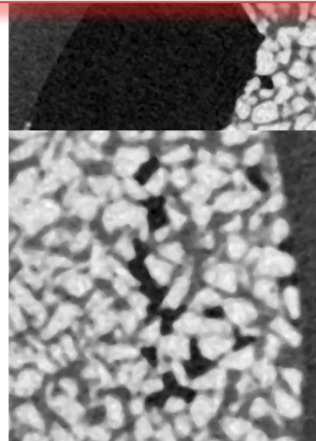
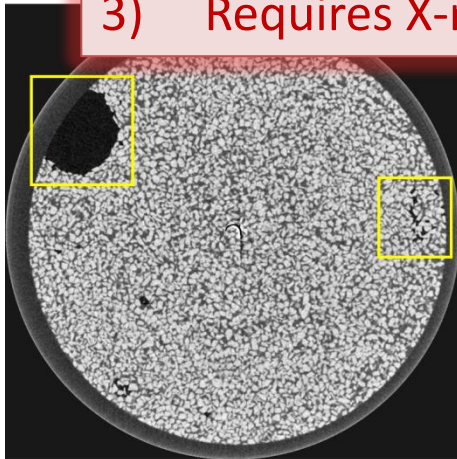


Multiphase flow

CT-scan



Sy

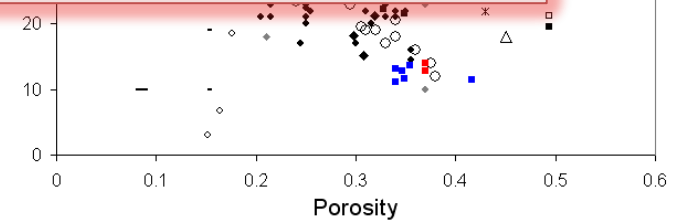


Objective: measure residual saturation
& relative permeability

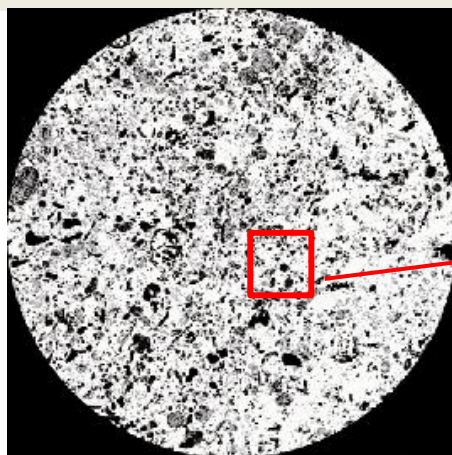


3 issues (at least)

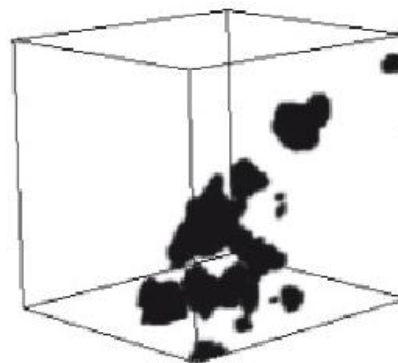
- 1) Experiments are usually long (1 – 2 days)
- 2) Problem with water (local radiolysis effects)
- 3) Requires X-ray transparent confinement cells



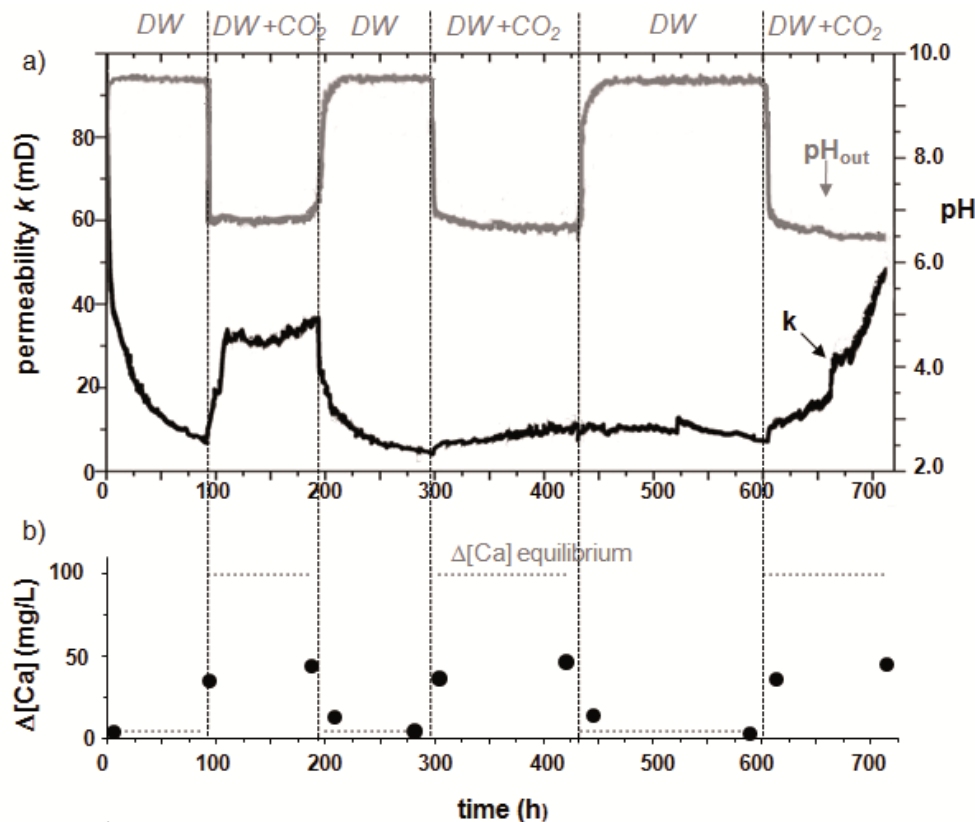
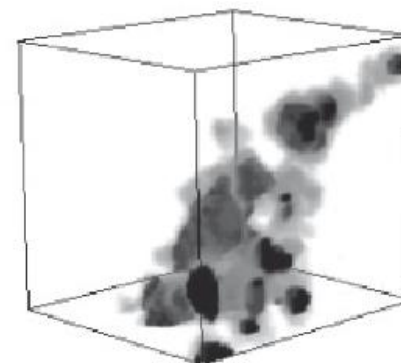
Dissolution and particles transport in carbonates



Pores > resolution



pores < resolution (μ -porosity)



Porosity always increases
(i.e. solid mass is removed)
but

Permeability may increase
or decrease depending on the
pH of the fluid.

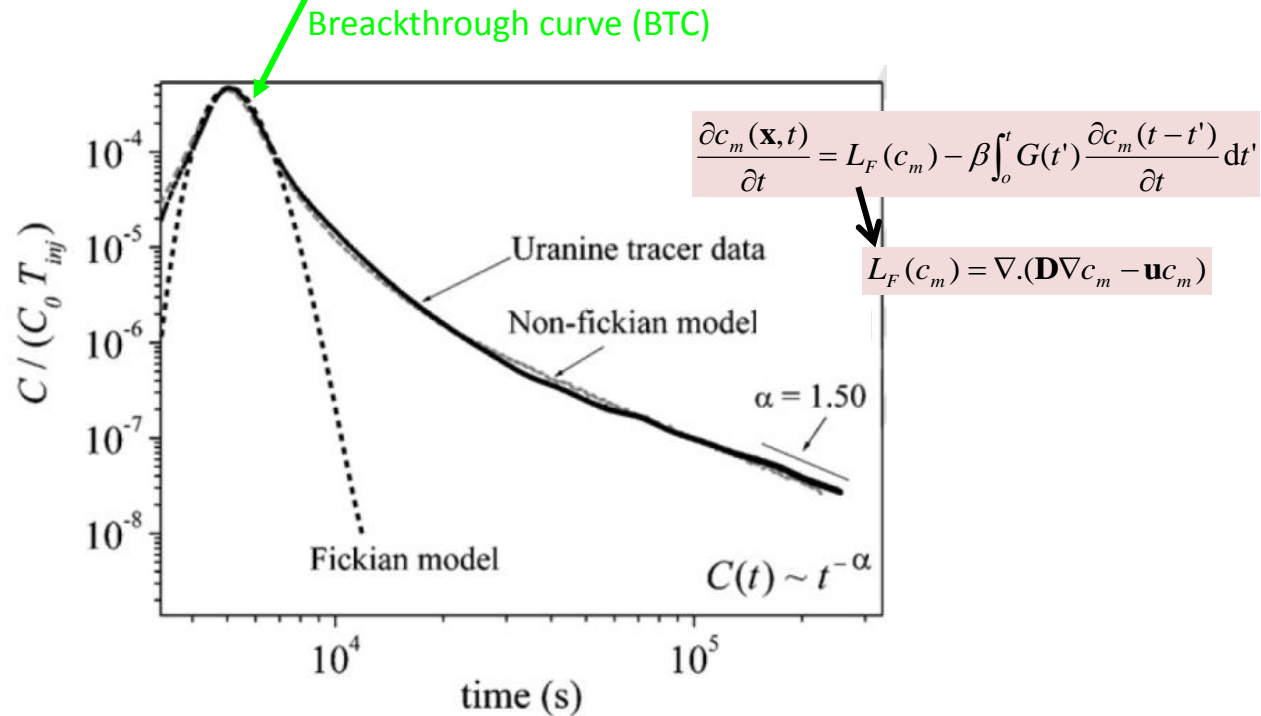
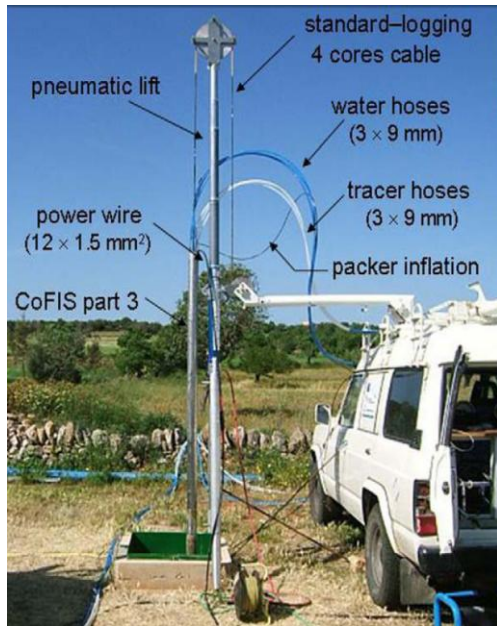
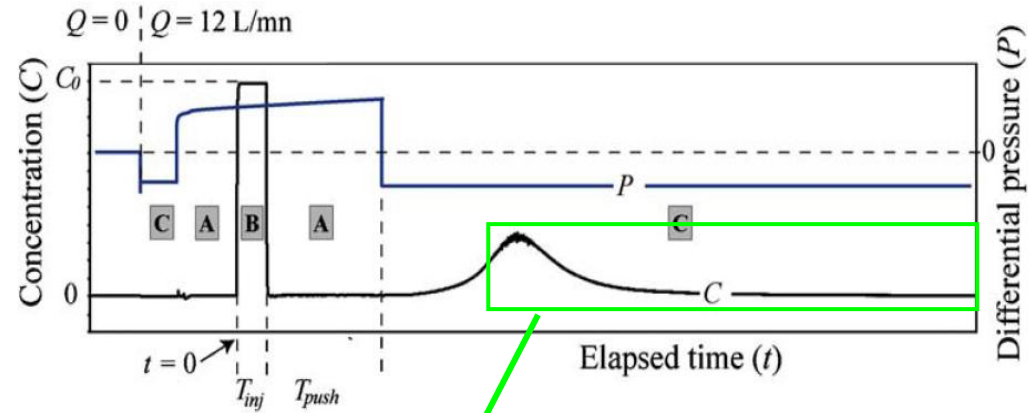
\Rightarrow Calcite μ -grains
detached and moved instead
of being dissolved

Applications to subsurface processes (dispersion of pollutants)

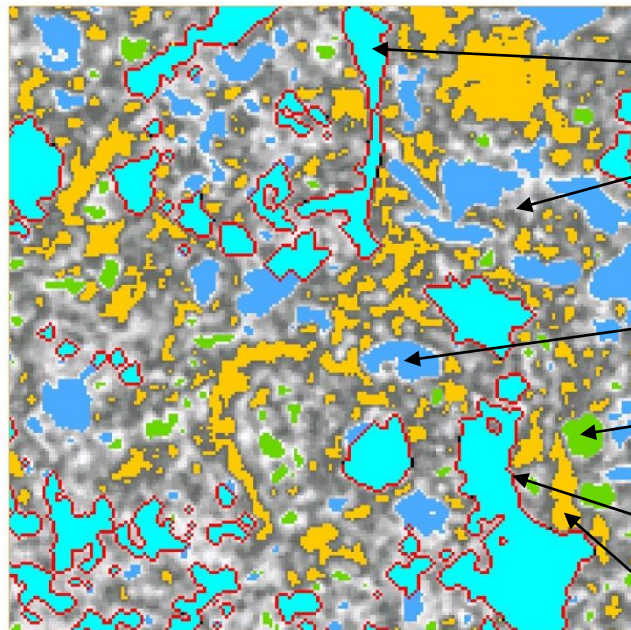
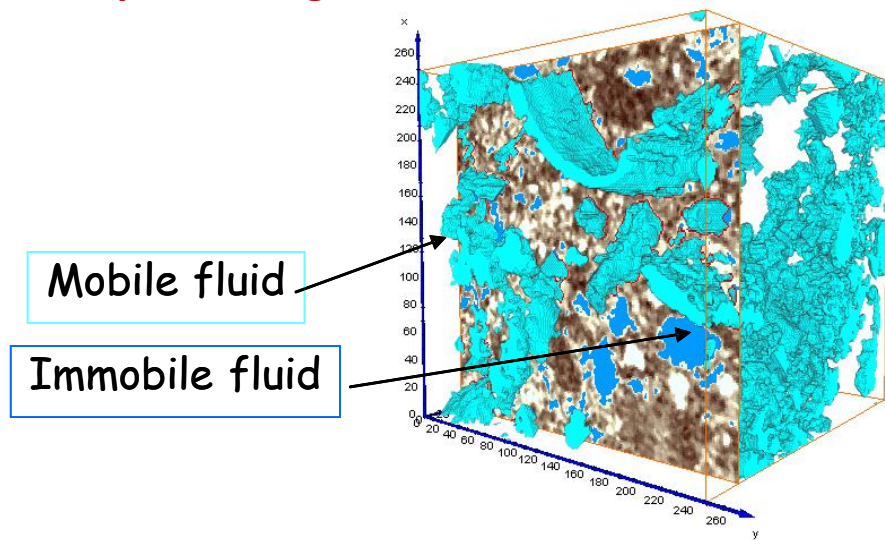
A main issue :

Explain the always-observed non-Fickian dispersion behavior
(while all models assume Fickian dispersion)

1) Field scale experiment



2) XRMT of the rock samples and specific data processing



Mobile domain (advection - dispersion)

Microporous matrix (diffusion $\phi d_0 / \tau$)
=> diffusion limited reaction

Macro dead-ends (diffusion d_0)

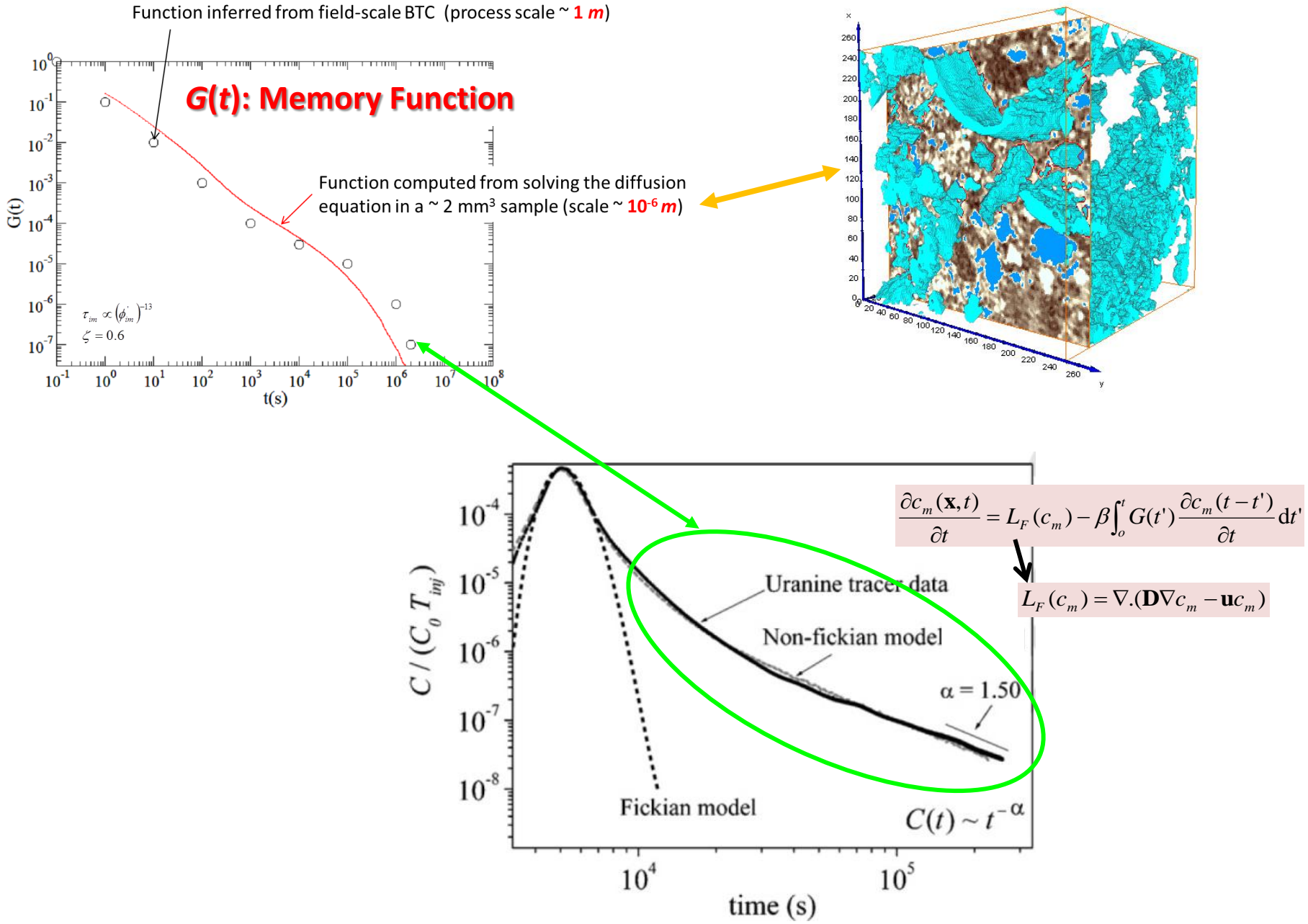
Trapped macroporosity (diffusion d_0)

Main reaction surface

Non diffusive fraction (solid)

Immobile domain

3) Compare $G(t)$ computed from the XRMT and $G(t)$ obtained from the field measurements



Main issues

Media composition (mineral and void distribution)

Media structure versus flow properties

Media structure versus solute transport & reactions

Structure of fluid-fluid interfaces (multiphase flow)

Media structure versus mechanical properties

Requirements (not all achievable so far, or at least not simultaneously ...)

3D !

Using samples of characteristic size around 1 cm.

Large range of scales (resolution from sub-100nm to tens of microns)

Fast imaging for allowing dynamic experiments

Resolving low contrast interfaces

Resolving mineral composition

If you think about the unique capabilities of SRX

what experiment comes immediately to your mind that you would like to do?

Static multi scale observations (zooming-in)

Dynamic experiments (monophasic and multiphasic fluid flow in porous media)

How should the sample environment look like?

What would be the demands on the environment from your samples?

For dynamic experiment, we need to build specific confinement cells (aluminum, carbon, ...?)

Which elements with absorption edges in the energy range covered by SRX (4.65keV to 22keV) would be of most interest for you?

Periodic table of elements. Elements of interest for SRX experiments are highlighted with red boxes:

- Period 4: K, Ca
- Period 5: Rb, Sr
- Period 6: Cs, Ba
- Period 6: Zr, Nb, Mo, Ru, Rh, Pd, Ag, Cd
- Period 6: In, Sn, Sb, Te, I, Xe
- Lanthanides (Period 7): La, Ce, Pr, Nd, Pm, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb
- Actinides (Period 8): Ac, Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, No

Legend for element information:

- atomic # →
- atomic symbol →
- English element name →
- common oxidation states →
- color →
- atomic mass (rounded) →

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Which elements with absorption edges in the energy range covered by SRX (4.65keV to 22keV) would be of most interest for you?

What would be more important for you for an experiment at SRX, very high spatial resolution (sub-100nm) or a large sample area (mm with sub-micron resolution)?

The ideal for us is from sub-100nm to 5 microns (using “zooming in”)

THANK YOU

